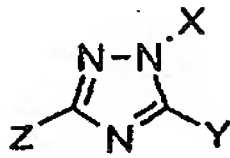




## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<b>(51) International Patent Classification <sup>7</sup> :</b> <b>C07D 401/04, 401/14, 409/14, 417/14, 213/83, A01N 43/653</b>	<b>A1</b>	<b>(11) International Publication Number:</b> <b>WO 00/24735</b> <b>(43) International Publication Date:</b> 4 May 2000 (04.05.00)
<b>(21) International Application Number:</b> PCT/US99/24751 <b>(22) International Filing Date:</b> 22 October 1999 (22.10.99) <b>(30) Priority Data:</b> 60/105,356                      23 October 1998 (23.10.98)                      US <b>(71) Applicant:</b> DOW AGROSCIENCES LLC [US/US]; 9330 Zionsville Road, Indianapolis, IN 46268 (US).		<b>(72) Inventors:</b> TISDELL, Francis, E.; 67 Parkview Road, Carmel, IN 46032 (US). JOHNSON, Peter, L.; 8016 North Dartmouth Road, Indianapolis, IN 46260 (US). PECHACEK, James, T.; 1230 Munsee Court, Indianapolis, IN 46260 (US). BIS, Scott, J.; 5267 Crenshaw Court, Carmel, IN 46033 (US). HEDGE, Vidyadhar, B.; 2532 Royal Boulevard, Carmel, IN 46032 (US). SCHOONOVER, Joe, R., Jr.; 7142 Marstella Drive, Brownsburg, IN 46112 (US). RIPA, Perry, V.; 1017 Sunfield Street, Sun Prairie, WI 53590 (US). DINTENFASS, Leonard, P.; 1150 East 57th Street, Indianapolis, IN 46220 (US). GIFFORD, James, M.; 800 N 200 E, Lebanon, IN 46052 (US). THIBAUT, Thomas, D.; 1520 North Franklin Road, Indianapolis, IN 46219 (US). ASH, Mary, L.; 762 Franklin Trace, Zionsville, IN 46077 (US). DEVRIES, Donald, H.; 9878 Woodlands Drive, Fishers, IN 46038 (US). MARTIN, Timothy, P.; 40 E. 87th Street, Indianapolis, IN 46240 (US). <b>(74) Agent:</b> STUART, Donald, R.; Dow AgroSciences LLC, 9330 Zionsville Road, Indianapolis, IN 46268 (US). <b>(81) Designated States:</b> AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG). <b>Published</b> <i>With international search report.</i>
<b>(54) Title:</b> INSECTICIDAL 1-(SUBSTITUTED PYRIDYL)-1,2,4-TRIAZOLES <div style="text-align: center; margin: 20px 0;">  <div style="display: inline-block; vertical-align: middle; margin-left: 20px;">(1)</div> </div> <b>(57) Abstract</b> <p>Compounds of formula (1) wherein one of X and Y is lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, or alkoxyalkyl; and the other of X and Y is optionally substituted phenyl, pyridyl, thienyl, cyclopropyl, or thiazolyl; and Z is substituted pyridyl are useful as insecticides and acaricides. New synthetic procedures and intermediates for preparing the compounds, pesticide compositions containing the compounds, and methods of controlling insects and mites using the compounds are also provided.</p>		

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## INSECTICIDAL 1-(SUBSTITUTED PYRIDYL)-1,2,4-TRIAZOLES

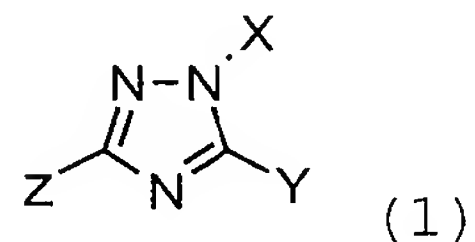
This invention provides new compounds that are useful as insecticides and acaricides, new synthetic procedures and intermediates for preparing the compounds, pesticide compositions containing the compounds, and methods of controlling insects and mites using the compounds.

There is an acute need for new insecticides and acaricides. Insects and mites are developing resistance to the insecticides and acaricides in current use. At least 400 species of arthropods are resistant to one or more insecticides. The development of resistance to some of the older insecticides, such as DDT, the carbamates, and the organophosphates, is well known. But resistance has even developed to some of the newer pyrethroid insecticides and acaricides. Therefore a need exists for new insecticides and acaricides, and particularly for compounds that have new or atypical modes of action.

A number of 3,5-diphenyl-1H-1,2,4-triazole derivatives have been described in the literature as having acaricidal activity. U.S. 5,482,951; JP 8092224, EP572142, JP 08283261. To applicants knowledge, however, none of these compounds has become a commercial product. Nitro furanyl triazoles are described by L.E. Benjamin and H.R. Snyder as antimicrobials (*J. Heterocyclic Chem.* **1976**, 13, 1115) and by others as antibacterials (*J. Med. Chem.* **1973**, 16(4), 312-319; *J. Med. Chem.* **1974**, 17(7), 756-758). The present invention provides novel compounds with commercial level activity against mites and insects.

This invention provides novel substituted pyridyl triazole derivatives especially useful for the control of insects and mites.

More specifically, the invention provides novel insecticidally active compounds of the formula (1)

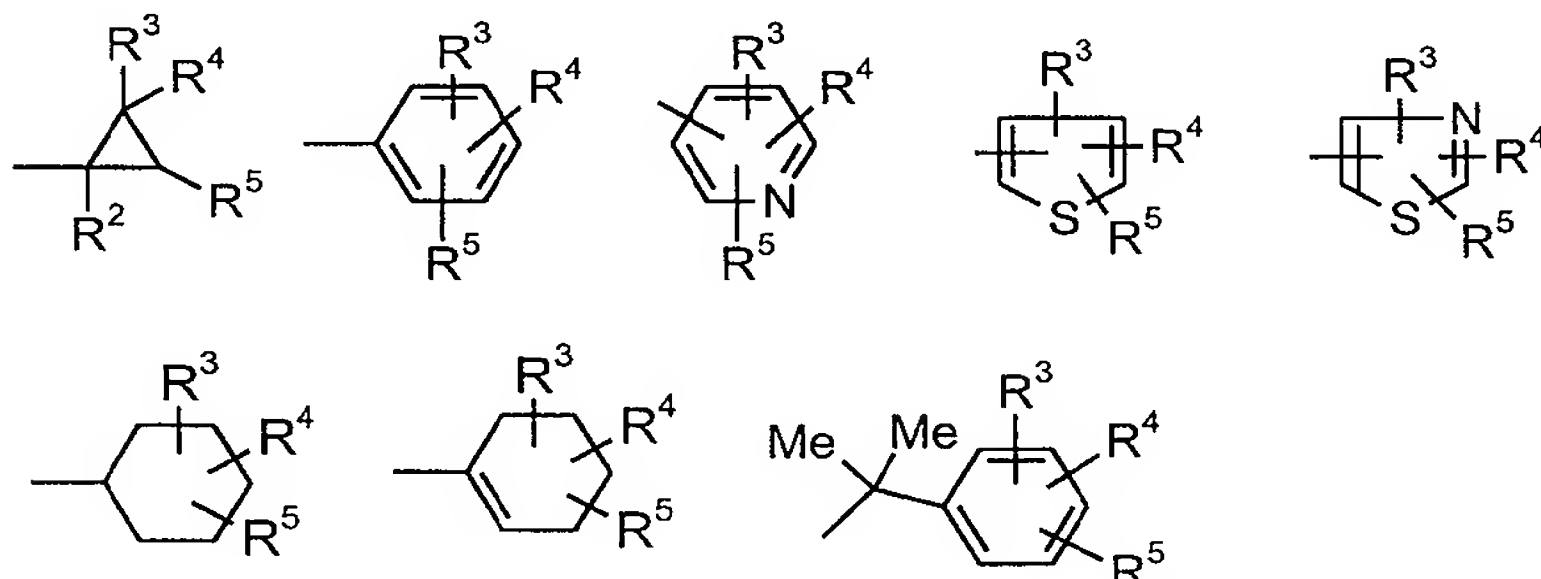


5 wherein

Z is pyridyl, optionally substituted with up to four groups independently selected from the group consisting of Cl, F, methyl, halomethyl, methoxy, halomethoxy, and methylthio;

10 one of X and Y is H, lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, alkoxyalkyl, phenyl, or substituted phenyl;

the other of X and Y is a group selected from

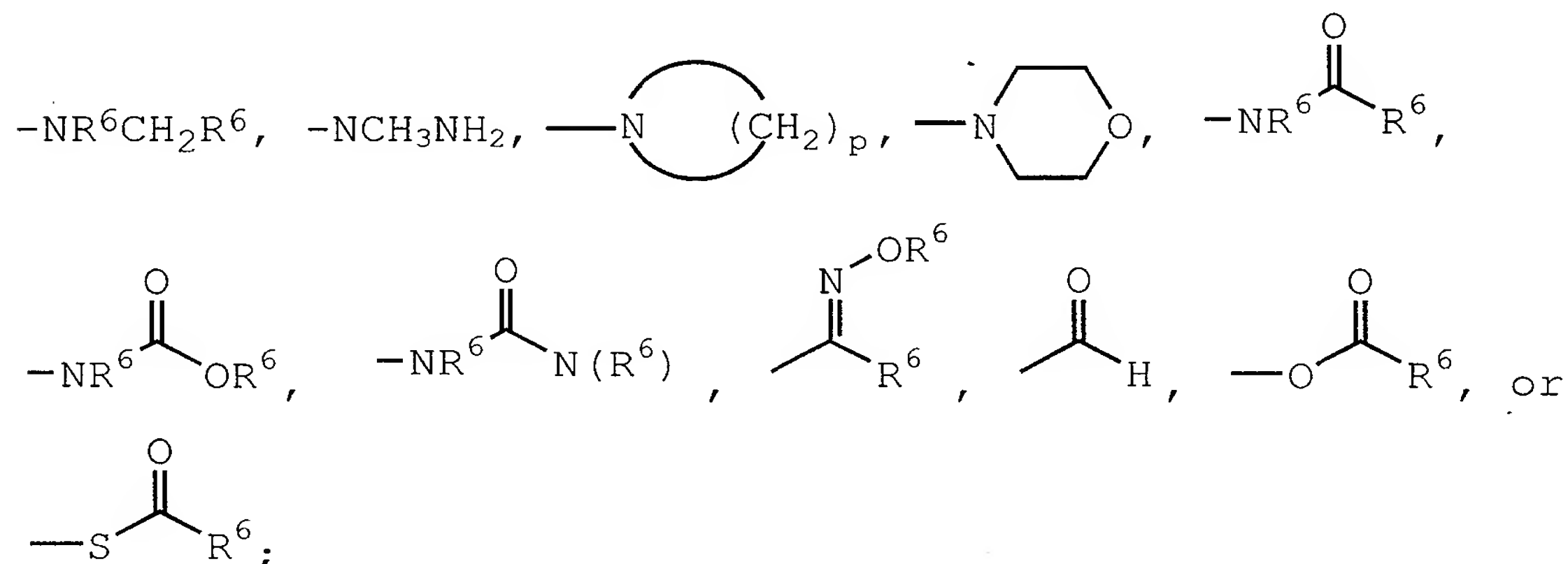


15 wherein

R<sup>2</sup> is halo, lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, or alkoxyalkyl;

R<sup>3</sup> is selected from H, halo, lower alkyl, (C<sub>7</sub>-C<sub>21</sub>) straight chain alkyl, hydroxy, lower alkoxy, haloalkyl, haloalkoxy, alkoxyalkyl, alkoxyalkoxy, lower alkenyl, 20 haloalkenyl, CN, NO<sub>2</sub>, CO<sub>2</sub>R<sup>6</sup>, CON(R<sup>6</sup>)<sub>2</sub>, (C<sub>3</sub>-C<sub>6</sub>) cycloalkyl, S(O)<sub>m</sub>R<sup>6</sup>, SCN, pyridyl, pyridyloxy, substituted pyridyl, substituted pyridyloxy, phenoxy, substituted phenoxy, isoxazolyl, substituted isoxazolyl, naphthyl, substituted

naphthyl, phenyl, substituted phenyl,  $-(CH_2)_nR^6$ ,  $-CH=CHR^6$ ,  
 $-C\equiv CR^6$ ,  $-CH_2OR^6$ ,  $-CH_2SR^6$ ,  $-CH_2NR^6R^6$ ,  $-OCH_2R^6$ ,  $-SCH_2R^6$ ,



$R^4$  and  $R^5$  are independently H, halo, lower alkyl,  
 lower alkoxy, haloalkyl, haloalkoxy, CN,  $NO_2$ ,  $CO_2R^6$ ,  
 $CON(R^6)_2$ , or  $S(O)_m$  alkyl, or

$R^4$  and  $R^5$  form a 5 or 6 member saturated or  
 10 unsaturated carbocyclic ring which may be substituted by  
 1 or 2 halo, lower alkyl, lower alkoxy or haloalkyl  
 groups;

$R^6$  is H, lower alkyl, haloalkyl, lower alkenyl,  
 lower alkynyl, phenyl, or substituted phenyl;

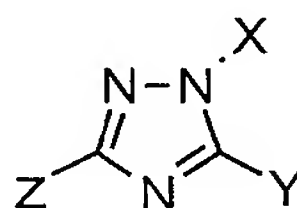
15  $m$  is 0, 1, or 2; and

$n$  is 1 or 2;

$p$  is an integer from 2 to 6;

or a phytologically acceptable acid addition salt or N-  
 oxide thereof.

20 The priority document described compounds of the  
 formula

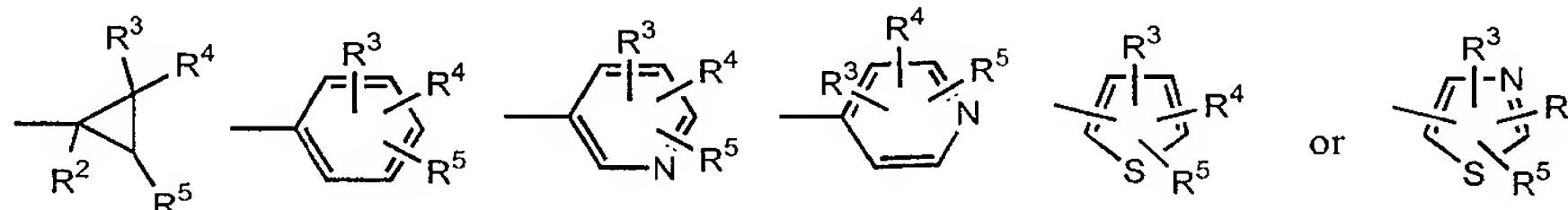


wherein

Z is 2-pyridyl, 3-pyridyl, or 4-pyridyl, optionally substituted with up to four groups independently selected from the group consisting of Cl, F, methyl, halomethyl, methoxy, and halomethoxy;

5 X is lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, or alkoxyalkyl;

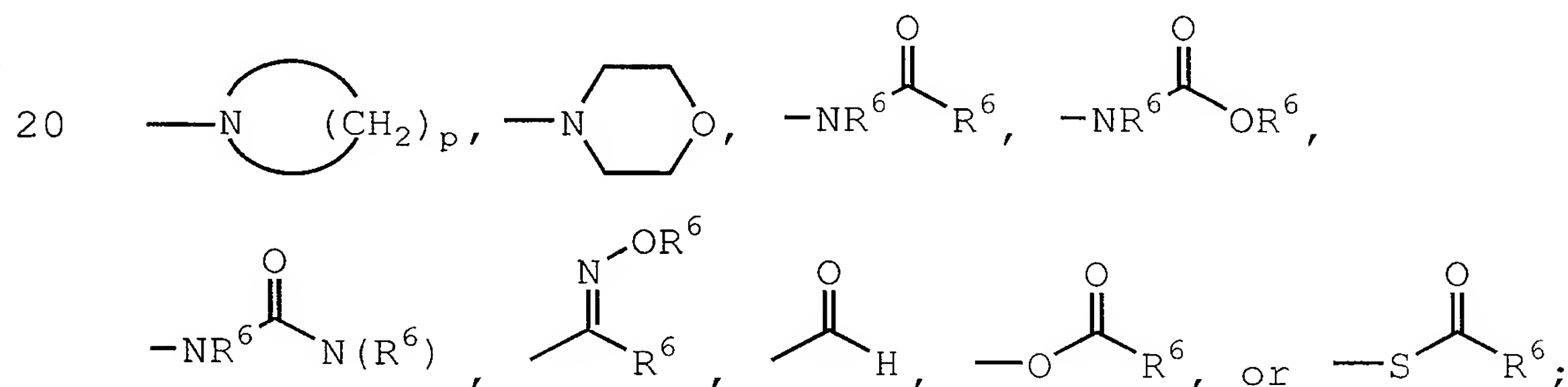
Y is a group selected from



wherein

10 R<sup>2</sup> is lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, or alkoxyalkyl;

R<sup>3</sup> is selected from H, halo, lower alkyl, (C<sub>7</sub>-C<sub>21</sub>) straight chain alkyl, hydroxy, lower alkoxy, haloalkyl, haloalkoxy, alkoxyalkyl, alkoxyalkoxy, lower alkenyl, haloalkenyl, CN, NO<sub>2</sub>, CO<sub>2</sub>R<sup>6</sup>, CON(R<sup>6</sup>)<sub>2</sub>, (C<sub>3</sub>-C<sub>6</sub>) cycloalkyl, S(O)<sub>m</sub>R<sup>6</sup>, SCN, pyridyl, substituted pyridyl, isoxazolyl, substituted isoxazolyl, naphthyl, substituted naphthyl, phenyl, substituted phenyl, -(CH<sub>2</sub>)<sub>n</sub>R<sup>6</sup>, -CH=CHR<sup>6</sup>, -C≡CR<sup>6</sup>, -CH<sub>2</sub>OR<sup>6</sup>, -CH<sub>2</sub>SR<sup>6</sup>, -CH<sub>2</sub>NR<sup>6</sup>R<sup>6</sup>, -OCH<sub>2</sub>R<sup>6</sup>, -SCH<sub>2</sub>R<sup>6</sup>, -NR<sup>6</sup>CH<sub>2</sub>R<sup>6</sup>,



R<sup>4</sup> and R<sup>5</sup> are independently H, halo, lower alkyl, lower alkoxy, haloalkyl, haloalkoxy, CN, CO<sub>2</sub>R<sup>6</sup>, CON(R<sup>6</sup>)<sub>2</sub>, or S(O)<sub>m</sub> alkyl, or

R<sup>4</sup> and R<sup>5</sup> form a 5 or 6 member saturated or unsaturated carbocyclic ring which may be substituted by 1 or 2 halo, lower alkyl, lower alkoxy or haloalkyl groups;

5 R<sup>6</sup> is H, lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, phenyl, or substituted phenyl;

m is 0, 1, or 2; and

n is 1 or 2;

p is an integer from 2 to 6;

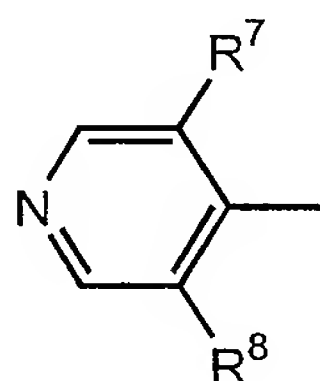
10 or a phytolegically acceptable acid addition salt or N-oxide thereof.

Preferred compounds of formula (1) include the following classes:

(1) Compounds of formula (1) wherein wherein one of  
15 X and Y is lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, substituted phenyl, or alkoxyalkyl; and the other of X and Y is optionally substituted phenyl, pyridyl, thienyl, cyclopropyl, or thiazolyl;

(2) Compounds of formula (1) wherein Z is a 4-  
20 pyridyl group optionally substituted with up to four groups independently selected from Cl, F, methyl, halomethyl, methylthio, methoxy, and halomethoxy.

(3) Compounds of class (2) wherein Z is a 4-pyridyl group of the formula



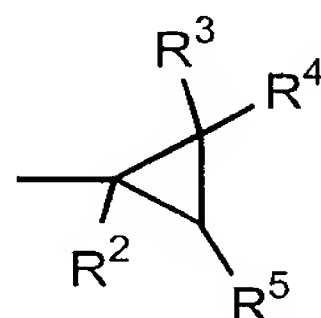
25 wherein R<sup>7</sup> and R<sup>8</sup> are independently H, Cl, F, methyl, halomethyl, methoxy, or halomethoxy.

(4) Compounds of class (3) wherein  $R^7$  and  $R^8$  are independently methyl, F or Cl.

(5) Compounds of class (4) wherein  $R^7$  and  $R^8$  are both F, where  $R^7$  is chloro and  $R^8$  is methyl, or where  $R^7$  is  
5 fluoro and  $R^8$  is methyl.

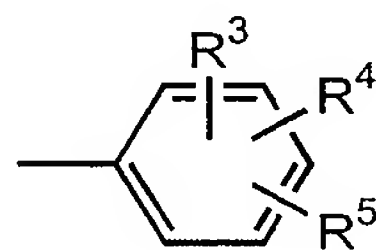
(6) Compounds of class (4) wherein  $R^7$  and  $R^8$  are both Cl.

(7) Compounds of formula (1), and particularly compounds of class (1), (2), (3), (4), (5), or (6) as  
10 defined above, wherein Y is a group of the formula



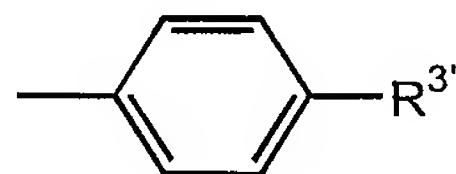
where  $R^2$ ,  $R^3$ ,  $R^4$ , and  $R^5$  are as defined in formula (1).

(8) Compounds of formula (1), and particularly compounds of class (1), (2), (3), (4), (5), or (6) as  
15 defined above, wherein Y is a group of the formula



where  $R^3$ ,  $R^4$ , and  $R^5$  are as defined in formula (1).

(9) Compounds of class (7) wherein Y is a group of the formula

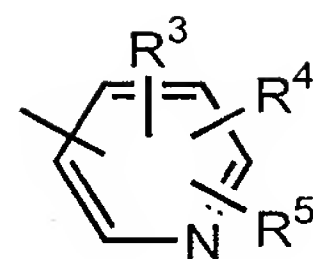


20

where  $R^{3'}$  is an electron withdrawing group such as Cl, Br,  $CF_3$  or  $NO_2$ .

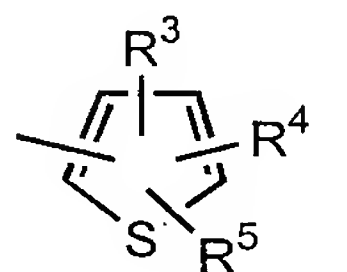
(10) Compounds of formula (1), and particularly compounds of class (1), (2), (3), (4), (5), or (6) as  
25 defined above, wherein Y is a group of the formula





where  $R^3$ ,  $R^4$ , and  $R^5$  are as defined in formula (1).

(11) Compounds of formula (1), and particularly compounds of class (1), (2), (3), (4), (5), or (6) as  
5 defined above, wherein Y is a group of the formula



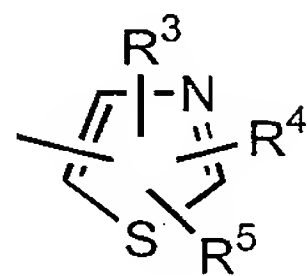
where  $R^3$ ,  $R^4$ , and  $R^5$  are as defined in formula (1).

(12) Compounds of class (11) wherein  $R^3$ ,  $R^4$ , and  $R^5$   
10 are independently selected from H, halo, alkyl, and alkoxy.

(13) Compounds of class (12) wherein  $R^3$ ,  $R^4$ , and  $R^5$   
are independently selected from H, alkyl, and halo.

(14) Compounds of class (13) wherein  $R^3$ ,  $R^4$ , and  $R^5$   
are independently selected from H, methyl, Cl, and Br.

15 (15) Compounds of formula (1), and particularly compounds of class (1), (2), (3), (4), (5), or (6) as defined above, wherein Y is a group of the formula



where  $R^3$ ,  $R^4$ , and  $R^5$  are as defined in formula (1).

20 (16) Compounds of formula (1) or any one of classes (1) through (15) wherein X is methyl.

The invention also provides new processes and intermediates for preparing compounds of formula (1) as well as new compositions and methods of use, which will be described in detail hereinafter.

5 Detailed Description Of The Invention

Throughout this document, all temperatures are given in degrees Celsius, and all percentages are weight percentages unless otherwise stated.

The term "lower alkyl" refers to (C<sub>1</sub>-C<sub>6</sub>) straight  
10 hydrocarbon chains and (C<sub>3</sub>-C<sub>6</sub>) branched and cyclic hydrocarbon groups.

The terms "lower alkenyl" and "lower alkynyl" refer to (C<sub>2</sub>-C<sub>6</sub>) straight hydrocarbon chains and (C<sub>3</sub>-C<sub>6</sub>)  
15 branched hydrocarbon groups containing at least one double or triple bond, respectively.

The term "lower alkoxy" refers to -O-lower alkyl.

The terms "halomethyl", "haloalkyl", and "haloalkenyl" refer to methyl, lower alkyl, and lower alkenyl groups substituted with one or more halo atoms.

20 The terms "halomethoxy" and "haloalkoxy" refer to methoxy and lower alkoxy groups substituted with one or more halo atoms.

The term "alkoxyalkyl" refers to a lower alkyl group substituted with a lower alkoxy group.

25 The term "alkoxyalkoxy" refers to a lower alkoxy group substituted with a lower alkoxy group.

The terms "substituted naphthyl", "substituted thienyl", "substituted pyrimidyl", "substituted pyrazolyl", "substituted pyridyl", and "substituted  
30 isoxaxolyl" refer to the ring system substituted with one or more groups independently selected from halo, halo

(C<sub>1</sub>-C<sub>4</sub>) alkyl, CN, NO<sub>2</sub>, (C<sub>1</sub>-C<sub>4</sub>) alkyl, (C<sub>3</sub>-C<sub>4</sub>) branched alkyl, phenyl, (C<sub>1</sub>-C<sub>4</sub>) alkoxy, or halo (C<sub>1</sub>-C<sub>4</sub>) alkoxy.

The term "substituted phenyl" refers to a phenyl group substituted with one or more groups independently selected from halo, (C<sub>1</sub>-C<sub>10</sub>) alkyl, branched (C<sub>3</sub>-C<sub>6</sub>) alkyl, halo (C<sub>1</sub>-C<sub>7</sub>) alkyl, hydroxy (C<sub>1</sub>-C<sub>7</sub>) alkyl, (C<sub>1</sub>-C<sub>7</sub>) alkoxy, halo (C<sub>1</sub>-C<sub>7</sub>) alkoxy, phenoxy, phenyl, NO<sub>2</sub>, OH, CN, (C<sub>1</sub>-C<sub>4</sub>) alkanoyl, benzoyl, (C<sub>1</sub>-C<sub>4</sub>) alkanoyloxy, (C<sub>1</sub>-C<sub>4</sub>) alkoxycarbonyl, phenoxycarbonyl, or benzoyloxy.

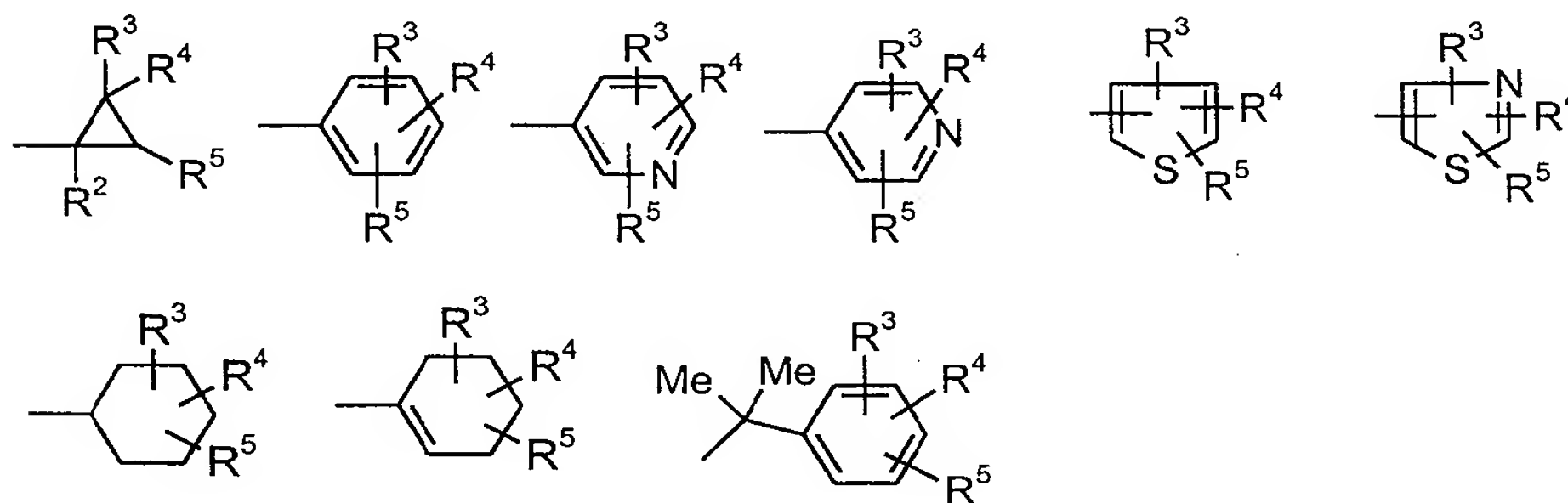
The term "pyridyl" refers to a 2-pyridyl, 3-pyridyl, or 4-pyridyl group.

Unless otherwise indicated, when it is stated that a group may be substituted with one or more substituents selected from an identified class, it is intended that the substituents may be independently selected from the class.

### Synthesis

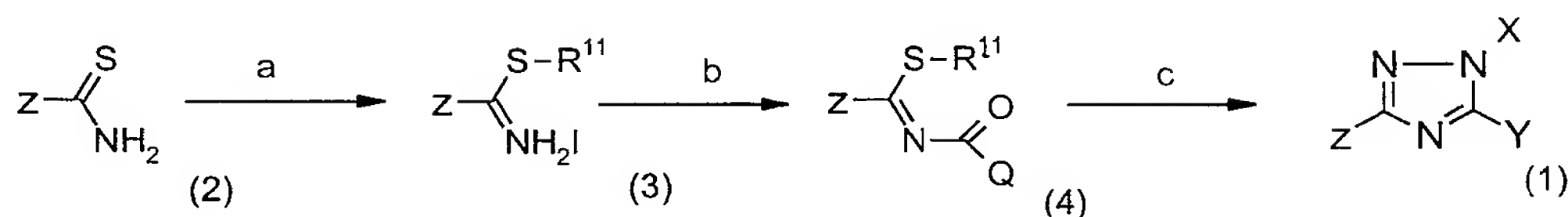
Compounds of formula (1) can be prepared by the methods described in U.S. Patent Nos. 5,380,944 and 5,284,860 (Production Methods 1, 2 and 3). Additional methods will be described hereinafter.

Compounds of formula (1) wherein X is H, lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, alkoxyalkyl, phenyl, or substituted phenyl; and Y is a group selected from



can be prepared by the method illustrated in Scheme I:

Scheme I



wherein X, Y and Z are as defined in formula (1) and R<sup>11</sup> is lower alkyl, preferably methyl.

The starting material of formula (2) used in Scheme I can be prepared by reacting a amide of formula Z-CONH<sub>2</sub> with phosphorus pentasulfide in pyridine at reflux.

In step **a** of Scheme I the compound of formula (2) is reacted with lower alkyl iodide, e.g. iodomethane, in acetone to provide a compound of formula (3). Acetone is the preferred solvent, however other polar aprotic solvents such as DMF or THF can be used.

In step **b** of Scheme I the compound of formula (3) is reacted with an acid chloride of formula Y-COCl in a nonreactive organic solvent such as benzene, toluene, xylenes, chloroform, dichloromethane, or 1,2-dichloroethane, at a temperature in the range from 0°C to the boiling point of the solvent.

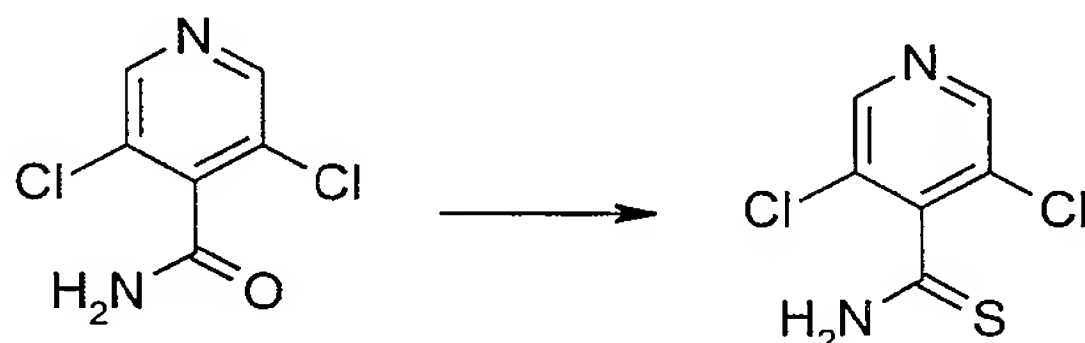
In step **c** of Scheme I, the N-acylimidate of formula (4) is reacted with an N-methylhydrazine to provide the product of formula (1). The reaction is carried out in benzene, toluene, xylenes, chloroform, dichloromethane, or 1,2-dichloroethane, at a temperature in the range from 0°C to the boiling point of the solvent.

In another of its aspects, the invention provides novel intermediates of the formulas (2), (3), and (4) as defined above.

## Example 1

S-methylthio-3,5-dichloro-4-pyridylimidinium iodide

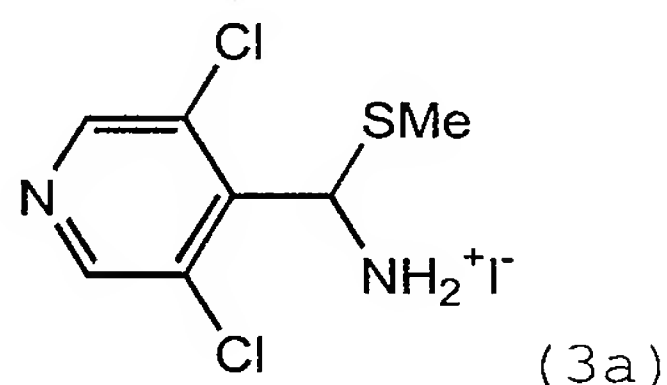
## A. Preparation of 3,5-dichloro-4-pyridinethioamide



5 Into a 3000-mL three-necked round bottom flask equipped with a condenser, mechanical stirrer under an atmosphere of nitrogen was added pyridine (1500 mL), then 3,5-dichloro-4-pyridine-carboxamide (92.9 g., 0.486 mole) (which dissolved), and tetraphosphorus decasulfide  
10 (237 g., 0.535 moles) (which had almost dissolved then a bright yellow precipitate formed and an exotherm heated the mixture to 60 °C). The slurry was allowed to stir for 1 hr (temperature had dropped to 45 °C) and then the temperature was then raised and when it reached 100 °C all  
15 of the solids had dissolved and continued heating to 118 °C and was maintained at 115 °C for 4 hr. The mixture was poured into water (3750 mL) carefully as gas began to evolve and the temperature of the aqueous solution rose to approximately 45 °C and was allowed to sit at room  
20 temperature over two nights. To the resulting mixture was added water (6000 mL) and was extracted with methylene chloride (3 X 2000 mL), washed with water (3 x 1000 mL) and the solvent removed in vacuo to give a brownish yellow liquid, with much pyridine present. The vacuum pump was  
25 connected to the rotary evaporator to remove the residual pyridine. The residue (brown solid) was triturated with diethyl ether (3 X 1500 mL), treated with decolorizing carbon and the solvent removed in vacuo to give a solid which contained pyridine. The yellow solid was slurried  
30 in water (2 X 200 mL) and dried in vacuo at 60 °C to give

63.2 g of a light yellow solid (62.8 % yield): mp 186-187 °C; TLC[50/50 ethyl acetate/hexanes] showed amide at  $R_f=0.31$  and thioamide  $R_f=0.53$ ;  $^1\text{H}$  NMR (DMSO- $d_6$ )  $\delta$  10.6 (s, b, 1H), 10.0 (s, b, 1H), 8.6 (s, 2H).

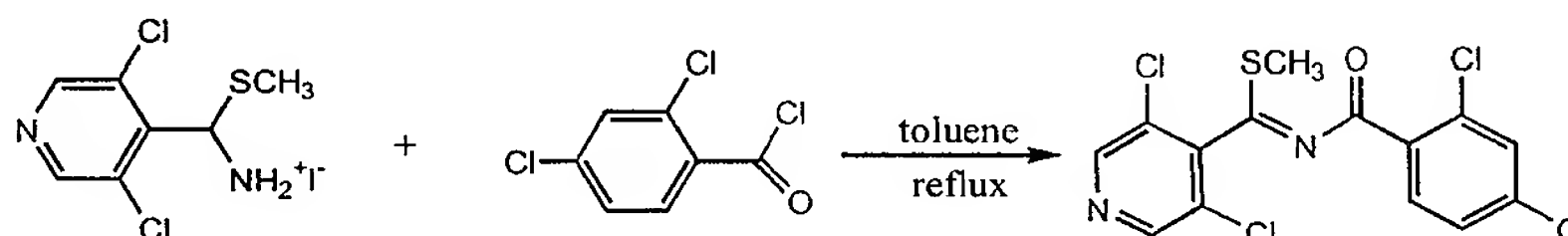
- 5 B. The following step illustrates the preparation of the S-methyl imidate of formula (3a)



Into a 3 L three necked flask equipped with a magnetic stirrer was added acetone (80mL) and 3,5-dichloro-4-pyridylthioamide (15.87 g, 76.6 mmol). To the stirred solution iodomethane (10.89 g, 4.77 mL, 76.6 mmol) was added dropwise. The slurry was stirred over night. The resulting yellow solids were removed via filtration and washed with ether to obtain 15.23 grams (57%) of S-methylthio-3,5-dichloro-4-pyridylimidinium iodide: mp 158-161 °C.  $^1\text{H}$  NMR (DMSO- $d_6$ )  $\delta$  8.8 (s, 2H), 7.8 (sb, 2H), 2.6 (s, 3H).

#### Example 2

20 N-(2,4-dichlorobenzoyl)-S-methylthio-3,5-dichloro-4-pyridylimidate

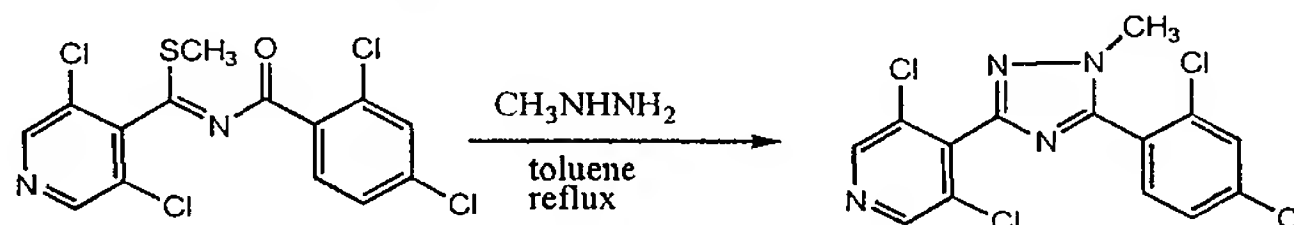


To a mixture of 2,4-dichlorobenzoyl chloride (0.94 g, 4.5 mmol) in 50 ml of dry toluene was added triethylamine (1.8 g, 18 mmol) and S-methylthio-3,5-dichloro-4-pyridylimidinium iodide (1.58 g, 4.5 mmol). The mixture was stirred at room temperature for two hours and then refluxed for two hours. The mixture was partitioned between brine and ether. The organic phase was dried

(MgSO<sub>4</sub>), the solvent evaporated, and the residue was chromatographed on silica gel using ethyl acetate/hexane as the eluant; 20:80. The product fractions were collected and evaporated to give 1.40 g (79% yield) of the title product as a reddish solid. Recrystallization afforded the title product (1.01 g., 57% yield) as a reddish solid. mp 104-105°C. <sup>1</sup>H NMR d 8.51 (s, 2H), 7.86-7.89 (d, 1H), 7.45 (s, 1H) 7.28-7.31 (m, 1H), 2.65 (s, 3H). Calculated for C<sub>14</sub>H<sub>8</sub>Cl<sub>4</sub>N<sub>2</sub>SO: C, 42.66; H, 2.05; N, 7.11; Found: C, 42.47 ; H, 1.96; N, 6.93.

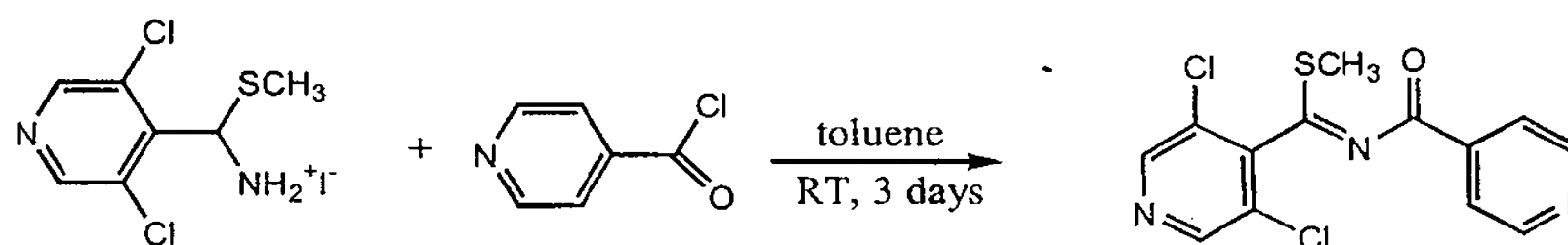
## Example 3

3-(3,5 dichloro-4-pyridyl)-5-(2,4-dichlorobenzyl)-1-methyl[1,2,4]triazole



A solution of 0.678 g (1.72 mmol) of the N-acyl-S-methylthioimide of Example 2 and 0.317 g (6.88 mmol) of methylhydrazine in 20 ml of toluene was refluxed four hours and then stirred overnight at ambient temperature. The solvent removed *in vacuo* and the residue was chromatographed on silica gel using ethyl acetate/hexane as the eluant; 25:75. The product fractions were collected and evaporated to give 0.31 g (48% yield) of the title product as white crystals. mp 79-81 °C. <sup>1</sup>H NMR d 8.6 (s, 2 H), 7.6 (s, 1H), 7.51-7.54 (d, 1H) 7.42-7.46 (m, 1H), 3.91 (s, 3H). Calculated for C<sub>14</sub>H<sub>8</sub>Cl<sub>4</sub>N<sub>4</sub>: C, 44.96; H, 2.16; N, 14.98; Found: C, 44.89 ; H, 2.10; N, 14.81.

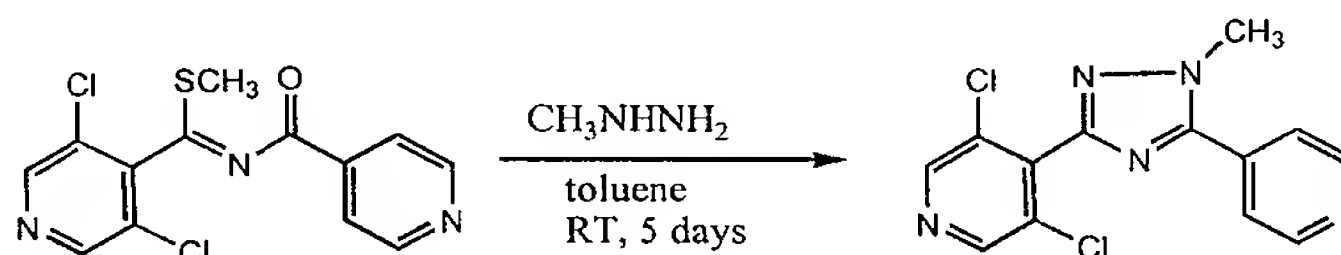
## Example 4

N-isonicotinoyl-S-methylthio-3,5-dichloro-4-pyridylimidate

- 5 To a mixture of isonicotinoyl chloride (1.14 g, 8.12 mmol) in 40 mL of dry toluene was added triethylamine (3.28 g, 32.5 mmol.) and S-methylthio-3,5-dichloropyridylimidinium iodide (2.83 g, 8.12 mmol). The mixture was stirred at room temperature for 72 hours.
- 10 The mixture was partitioned between brine and ether. The organic phase was dried (MgSO<sub>4</sub>), the solvent evaporated, and the residue was chromatographed on silica gel using ethyl acetate/hexane as the eluant; 50:50. The product fractions were collected and evaporated to give 0.88 g
- 15 (33% yield) of the title product as tan solid. mp 122-123 °C. <sup>1</sup>H NMR d 8.79-8.81 (d, 2H), 8.52 (s, 2H), 7.82-7.84 (d, 2H), 2.69 (s, 3H); Calculated for C<sub>13</sub>H<sub>9</sub>Cl<sub>2</sub>N<sub>3</sub>SO: C, 47.89; H, 2.79; N, 12.88; Found: C, 47.74 ; H, 2.69; N, 12.63.

20

## Example 5

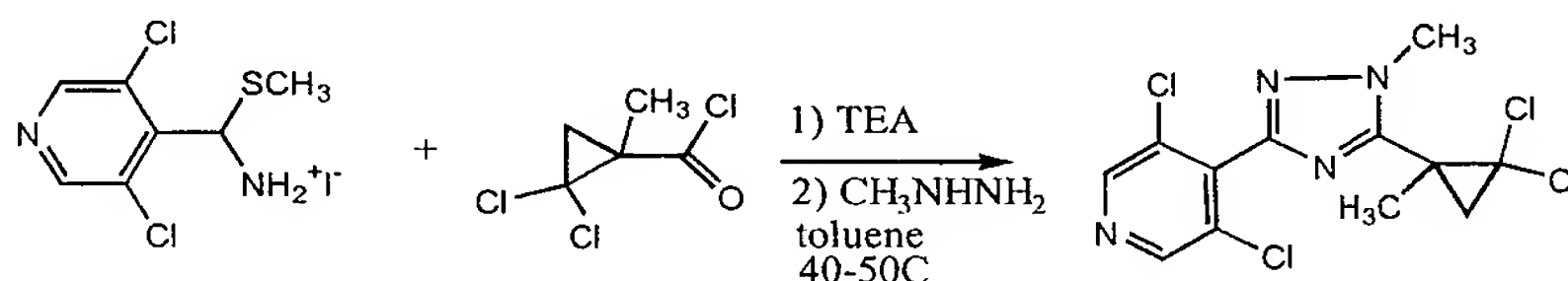
3-(3,5 dichloro-4-pyridyl)-5-isonicotinyl-1-methyl[1,2,4]triazole

- 25 A solution of 0.55 g (1.69 mmol) of the N-acyl-S-methylthioimidate of Example 4 and 0.155 g (3.37 mmol) of methylhydrazine in 10 ml of toluene was stirred 5 days at ambient temperature. The solvent was removed *in vacuo* and the residue was chromatographed on silica gel using
- 30 ethyl acetate as the eluant. The product fractions were collected and evaporated to give 0.222 g (43% yield) of



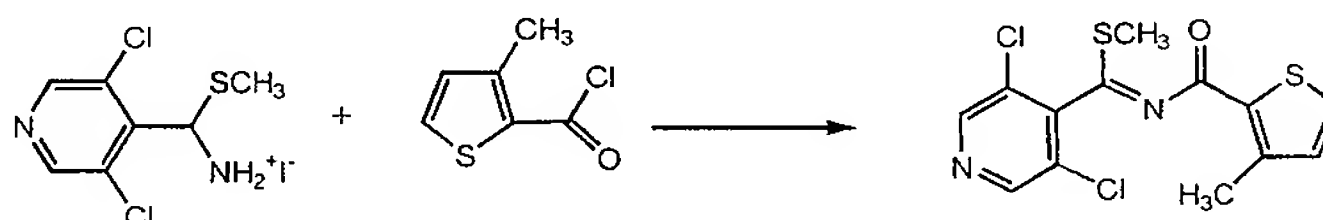
the title product as white crystals. mp 130-132 °C. <sup>1</sup>H NMR d 8.84-8.86 (d, 2 H), 8.64 (s, 2H), 7.72-7.74 (d, 2H), 4.20 (s, 3H). Calculated for C<sub>13</sub>H<sub>9</sub>Cl<sub>2</sub>N<sub>5</sub>: C, 51.00; H, 2.97; N, 22.88; Found: C, 50.55 ; H, 3.04; N, 22.08.

5

Example 63-(3,5 dichloro-4-pyridyl)-5-(2,2-dichloro-1-methylcyclopropyl)-1-methyl[1,2,4]triazole

10 To a solution of S-methylthio-3,5-dichloropyridylidinium iodide ( 0.585 g, 1.67 mmol) and triethylamine (0.675 g, 6.68 mmol.) in 15 mL of dry toluene was added a solution of 2,2-dichloro-1-methylcyclopropanoyl chloride (0.337 g, 1.8 mmol) in 10 ml  
15 toluene. The mixture was refluxed three hours. A solution of methylhydrazine (0.317 g , 6.88 mmol) in 10 mL of toluene was added dropwise and the mixture refluxed one hour and then stirred overnight at ambient temperature. The solvent removed *in vacuo* and the  
20 residue was chromatographed on silica gel using ethyl acetate/hexane as the eluant; 25:75. The product fractions were collected and evaporated to give 79 mg (13% yield) of the title product as yellow solid. mp 79-83 °C.

25

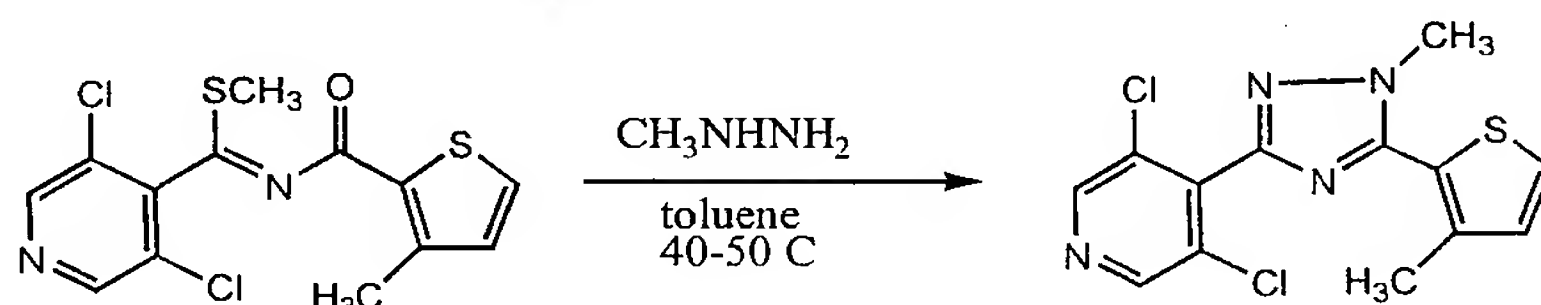
Example 7N-(3-Methyl-2-thienoyl)-S-methylthio-3,5-dichloro-4-pyridylidinate

30 Pyridine (0.51 ml, 6.3 mmol) was added dropwise to a slurry of 3-methyl-2-thiophenecarbonyl chloride (0.48g,

3.0 mmol) and S-methylthio-3,5-dichloropyridylimidinium iodide (1.05 g, 3.0 mmol) in 5 mL of 1,2-dichloroethane, under N<sub>2</sub>, at room temperature. After stirring at room temperature for 60 minutes the reaction mixture was  
 5 poured into H<sub>2</sub>O (25 ml) and extracted with ethyl acetate (3 X 25 ml). The combined organic extracts were washed with H<sub>2</sub>O (1 X 25 ml), saturated sodium chloride (1 X 25 ml), dried over anhydrous MgSO<sub>4</sub>, filtered and concentrated *in vacuo* to give 0.99 g of a yellow oil. This was  
 10 chromatographed over silica gel (MPLC), eluting with 90% hexane / 10% ethyl acetate. Isolation of the major product gave 0.827 g (80% yield) of the title compound as a faint yellow solid: mp 99-101°C. <sup>1</sup>H NMR CDCl<sub>3</sub> δ 8.51 (s, 2H), 7.45 (d, 1H), 6.94 (d, 1H), 2.64 (s, 3H), 2.49  
 15 (s, 3H).

## Example 8

1-Methyl-3-(3,5-dichloro-4-pyridyl)-5-(3-methyl-2-thienyl)[1,2,4]triazole

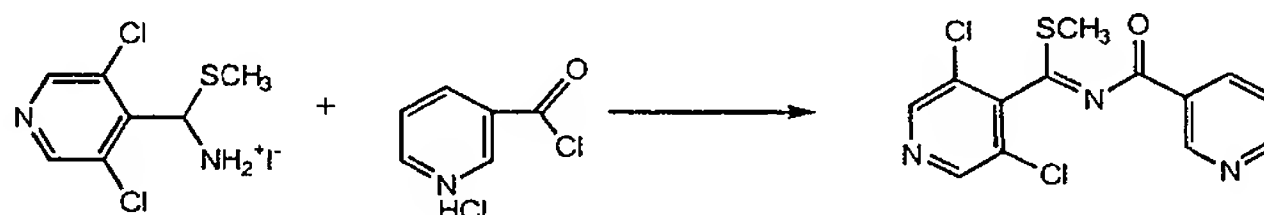


20 Methylhydrazine (0.225 ml, 4.2 mmol) was added dropwise to a solution of the N-acyl-S-methylthioimide of Example 7 (0.725 g, 2.1 mmol) in 5 ml of toluene, under N<sub>2</sub>, at room temperature. After stirring at room temperature for 24 hours, TLC analysis showed a 2:1  
 25 mixture of starting material to product. An additional 0.2 ml of methylhydrazine was added and the mixture warmed to 40°C. After 5 hours TLC shows a 1:1 mixture of starting material to product. An additional 0.2 ml of methylhydrazine was added and stirring continued for 24  
 30 hours at 40-50°C at which time TLC analysis indicated that all of the starting material had been consumed. The reaction mixture was concentrated *in vacuo* and the

resultant yellow oil was chromatographed over silica gel (MPLC), eluting with 80% hexane / 20% ethyl acetate. Isolation of the major product gave 0.422 g (65% yield) of the title compound as a faint yellow oil.  $^1\text{H}$  NMR  $\text{CDCl}_3$

5 8.61(s, 2H), 7.47(d, 1H), 7.02(d, 1H), 4.05(s, 3H), 2.40(s, 3H).

## Example 9

N-3-pyridoyl-S-methylthio-3,5-dichloro-4-pyridylimidate

10

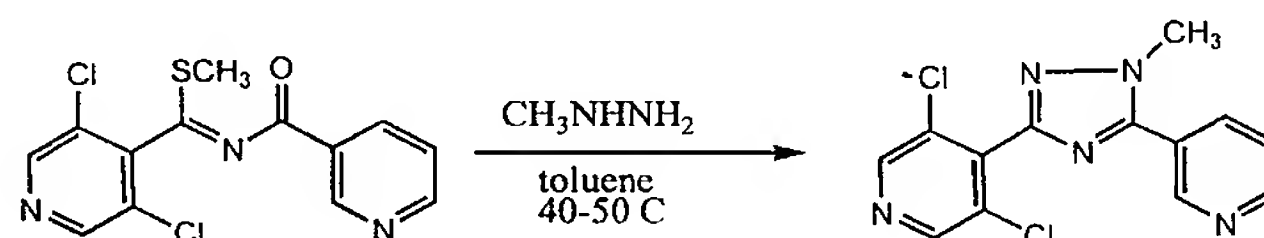
Pyridine (0.78 ml, 9.6 mmol) was added dropwise to a mixture of S-methylthio-3,5-dichloropyridylimidinium iodide (1.05 g, 3.0 mmol) and nicotinyl chloride hydrochloride (0.53 g, 3.0 mmol) in 5 ml of 1,2-

15 dichloroethane, under  $\text{N}_2$ , at room temperature. After 90 minutes at room temperature the reaction mixture was poured into  $\text{H}_2\text{O}$  (25 ml) and extracted with ethyl acetate (3 x 25 ml). The combined organic extracts were washed with  $\text{H}_2\text{O}$  (1 X 25 ml), saturated sodium chloride (1 X 25

20 ml), dried over anhydrous  $\text{Na}_2\text{SO}_4$ , filtered and concentrated *in vacuo* to give 0.76 g of a brown oil. This was chromatographed on silica gel (MPLC) eluting with 70% hexane / 30% ethyl acetate. Isolation of the major product gave 0.714 g (73% yield) of the desired

25 product as a yellow oil which slowly solidified: mp 106-108°C.  $^1\text{H}$  NMR  $\text{CDCl}_3$  9.27 (d, 1H), 8.78 (dd, 1H), 8.51 (s, 2H), 8.28 (m, 1H) 7.40 (m, 1H), 2.68 (s, 3H).

## Example 10

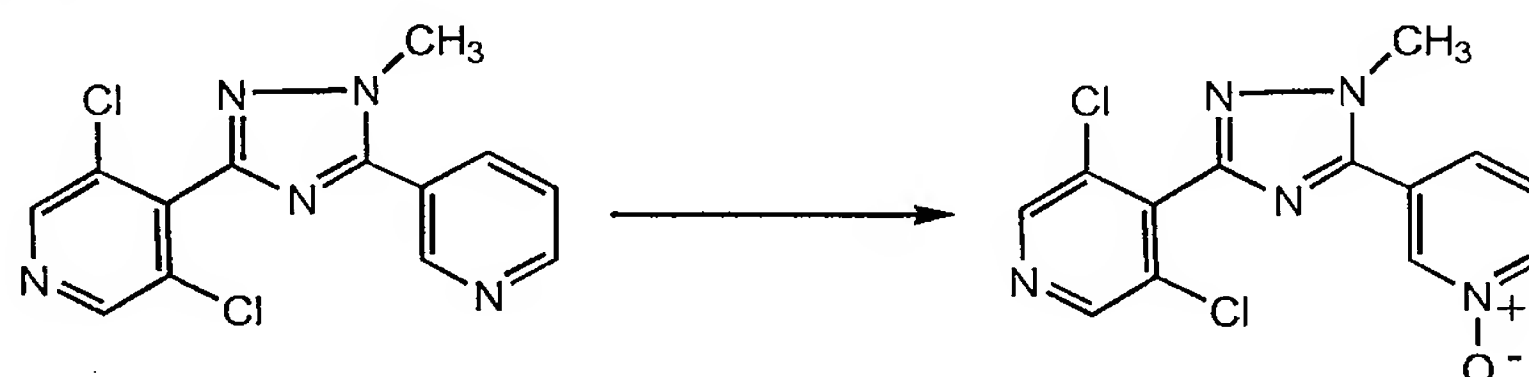
1-Methyl-3-(3,5-dichloro-4-pyridyl)-5-(3-pyridyl)[1,2,4]triazole

5 Methylhydrazine (0.20 ml, 3.8 mmol) was added to a solution of the N-acyl-S-methylthioimide of Example 9 (0.606 g, 1.9 mmol) in 5 ml of toluene, under N<sub>2</sub>, at room temperature. The resultant mixture was warmed to ~50°C. After stirring at 60-70°C for five hours TLC analysis

10 showed only a trace amount of the starting material present. The reaction mixture was concentrated in vacuo. The residual yellow oil was chromatographed on silica gel (MPLC), eluting with 50% hexane / 50% ethyl acetate. Isolation of the major product gave 0.291g (50% yield) of

15 the desired product as a yellow oil. <sup>1</sup>H NMR CDCl<sub>3</sub> 9.05 (d, 1H), 8.80 (dd, 1H), 8.63 (s, 2H), 8.14 (m, 1H), 7.51 (m, 1H), 4.17 (s, 3H).

## Example 11

1-Methyl-3-(3,5-dichloro-4-pyridyl)-5-(3-pyridyl-N-oxide)[1,2,4]triazole

A solution of the pyridine derivative of Example 10 (0.150 g, 0.5 mmol) in 2 ml of dichloromethane was treated in one portion with *m*-chloroperoxybenzoic acid

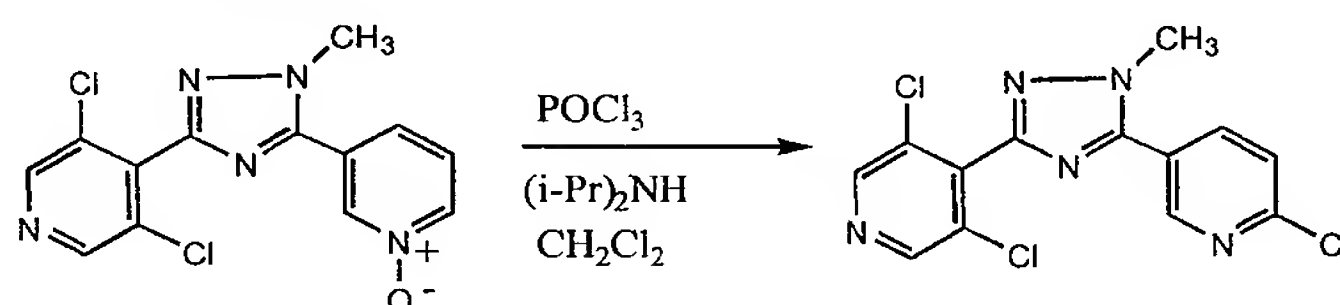
25 (0.173 g, ~0.55 mmol, 50-60%), while cooling in an ice bath. The resultant mixture was gradually allowed to warm to room temperature. After five hours at room temperature TLC analysis indicated that all of the starting material had been consumed. The reaction

30 mixture was diluted with dichloromethane (25 ml), washed

with 2N NaOH (2 X 10 ml) and saturated sodium chloride (1 X 10 ml). The organic phase was dried over anhydrous MgSO<sub>4</sub>, filtered and concentrated *in vacuo* to give 0.105 g (65% yield) of the desired product as a light yellow foam. <sup>1</sup>H NMR CDCl<sub>3</sub> 8.63 (d, 3H), 8.35 (m, 1H), 7.70 (d, 1H), 7.47 (t, 1H), 4.18 (s, 3H).

## Example 12

1-Methyl-3-(3,5-dichloro-4-pyridyl)-5-(6-chloro-3-pyridyl)[1,2,4]triazole



10

Phosphorus oxychloride (48 ml, 0.52 mmol) and diisopropylamine (73 ml, 0.52 mmol) were added simultaneously, via syringe, to a solution of the pyridine-N-oxide of Example 11 (83 mg, 0.26 mmol) in 2 ml of dichloromethane, under N<sub>2</sub>, at room temperature. After two hours TLC analysis showed much starting material still present and two minor products. The reaction mixture was concentrated *in vacuo* and the residue taken up in POCl<sub>3</sub> (2 ml) and heated to reflux. After two hours it appeared that all of the starting material had been consumed. The reaction mixture was cautiously poured into 2N NaOH (10 ml) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 X 10 mL). The combined organic extracts were washed with saturated sodium chloride (1 X 10 ml), dried over anhydrous MgSO<sub>4</sub>, filtered and concentrated *in vacuo* to give 45 mg of a yellow oil. This was chromatographed on silica gel (MPLC), eluting with 60% hexane / 40% ethyl acetate. Isolation of the major product gave 17 mg (19% yield) of the desired product as a yellow oil. <sup>1</sup>H NMR CDCl<sub>3</sub> 8.83 (d, 1H), 8.64 (sb, 2H), 8.11-8.14 (dd, 1H), 7.53-7.56 (d, 1H), 4.16 (s, 3H).

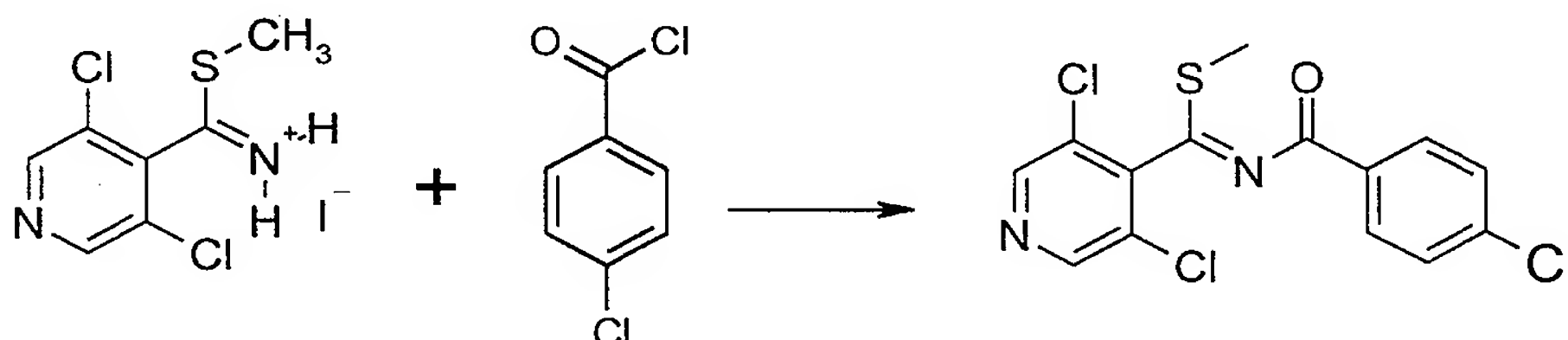
30

## Example 13

3-(3,5 dichloro-4-pyridyl N-oxide)-5-(4-chlorophenyl)-1-methyl[1,2,4]triazole

A solution of 0.378 g (1.12 mmol) of 3-(3,5 dichloro-4-pyridyl)-5-(4-chlorophenyl)-1-methyl[1,2,4]triazole and 0.414 g (1.2 mmol) of meta-chloroperbenzoic acid in 12 ml of methylene chloride was stirred 3 days at ambient temperature. The solvent removed *in vacuo* and the residue was chromatographed on silica gel using ethyl acetate as the eluant. The product fractions were collected and evaporated to give 0.33 g (83% yield) of the title product as a yellow solid. mp 161-166 °C. <sup>1</sup>H NMR d 8.28 (s, 2 H), 7.71-7.74 (d, 2H), 7.53-7.56 (d, 2H), 4.12 (s, 3H).

## Example 14

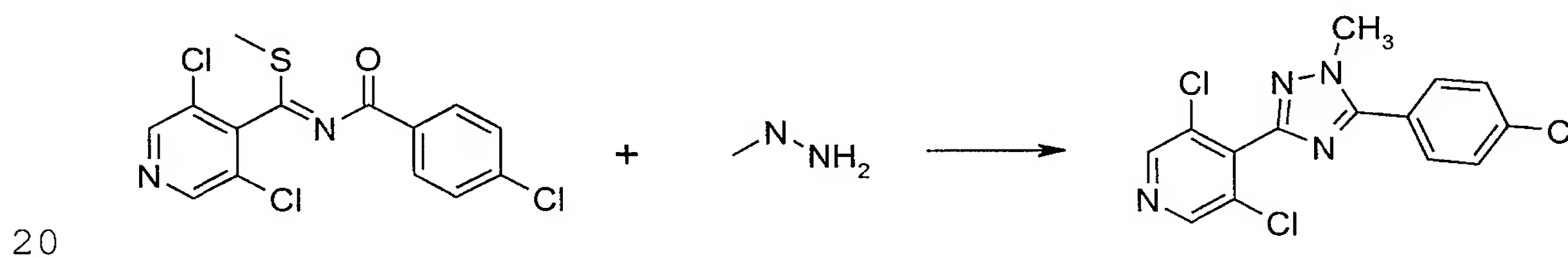
Preparation of N-p-chlorobenzoyl-3,5-dichloropyridin-4-yl methyl thioimide

Into a 1000 mL three necked round bottom flask equipped with a mechanical stirrer, thermometer and condenser under an atmosphere of nitrogen was added methyl 3,5-dichloropyridin-4-yl thioimide (63.4 g, 0.163 mol, 90% pure), 1,2-dichloroethane (327 mL) and pyridine (25.8 g, 26.3 mL, 0.326 mol). The temperature of the stirred mixture was raised to 45 °C and dropwise was added p-chlorobenzoyl chloride (28.6 g, 20.8 mL, 0.163 mol) over a 20-min period. An exotherm raised the temperature to 66 °C and the slurry was allowed to stir at that temperature for 0.5 h. The slurry was cooled to room

temperature and the solids removed via filtration and washed with a small portion of EDC. An equal volume of methylene chloride was added to the filtrate which was then washed with dilute acid (2 x 200 mL), saturated sodium bicarbonate, brine, dried ( $\text{Na}_2\text{SO}_4$ ), and the solvent removed in vacuo to give 67 g. The crude material was put through a plug (670 g) of silica gel eluting with methylene chloride which removed all of the polar material which stayed at the origin and removed the non-polar by taking appropriate fractions. Total clean product of 48.2 g (81.8% yield) as yellow crystalline material: mp 121-122 °C;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  8.5 (s, 2H), 8.0 (d,  $J=8.7$  Hz, 2H), 7.4 (d,  $J=8.7$  Hz, 2H), 2.6 (s, 3H). Anal. Calcd. for  $\text{C}_{14}\text{H}_9\text{Cl}_3\text{N}_2\text{O}$ : C, 46.75; H, 2.52; N, 7.79. Found: C, 46.75; H, 2.51; N, 7.67.

## Example 15

Preparation of 3-(3,5-dichloro-4-pyridyl)-5-(4-chlorobenzyl)-1-methyl[1,2,4]triazole



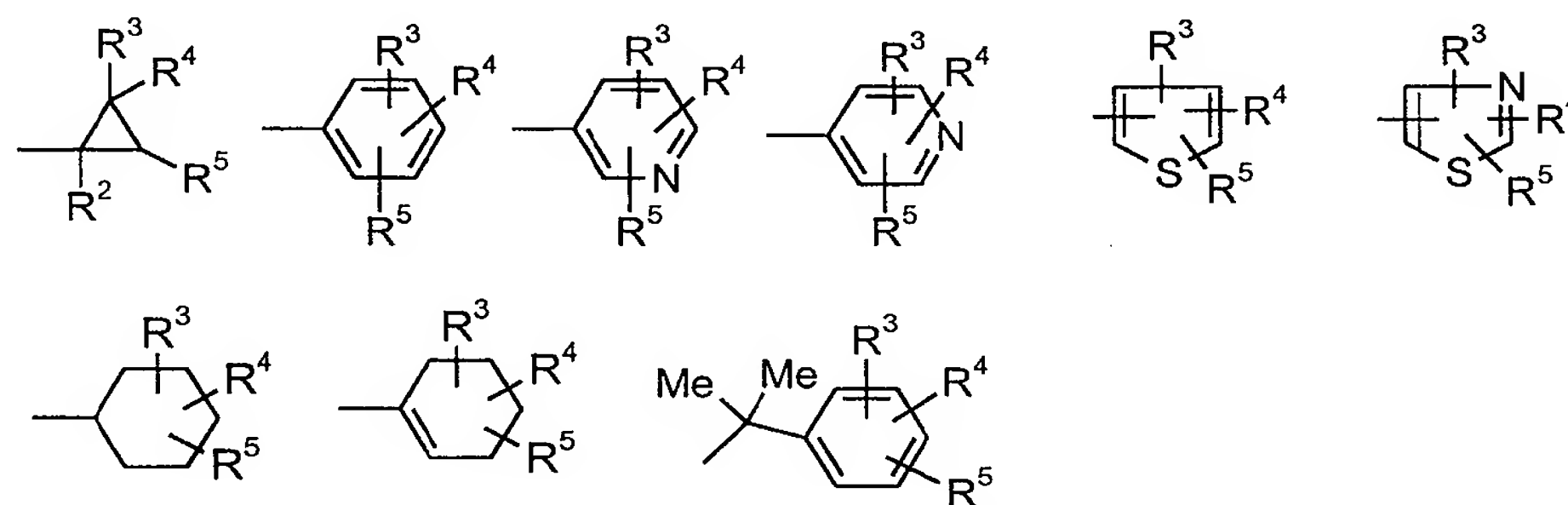


the solvent removed in vacuo to give an oil which crystallized on standing. The solids were dissolved in refluxing absolute ethanol (200 mL) and the hot solution seeded with an authentic sample of the title compound.

- 5 The resulting solids were removed via filtration and dried in vacuo at 70 °C to give 31.6 g (52% yield) of the title compound: mp 142-143 °C; 98.9% pure by GC analysis

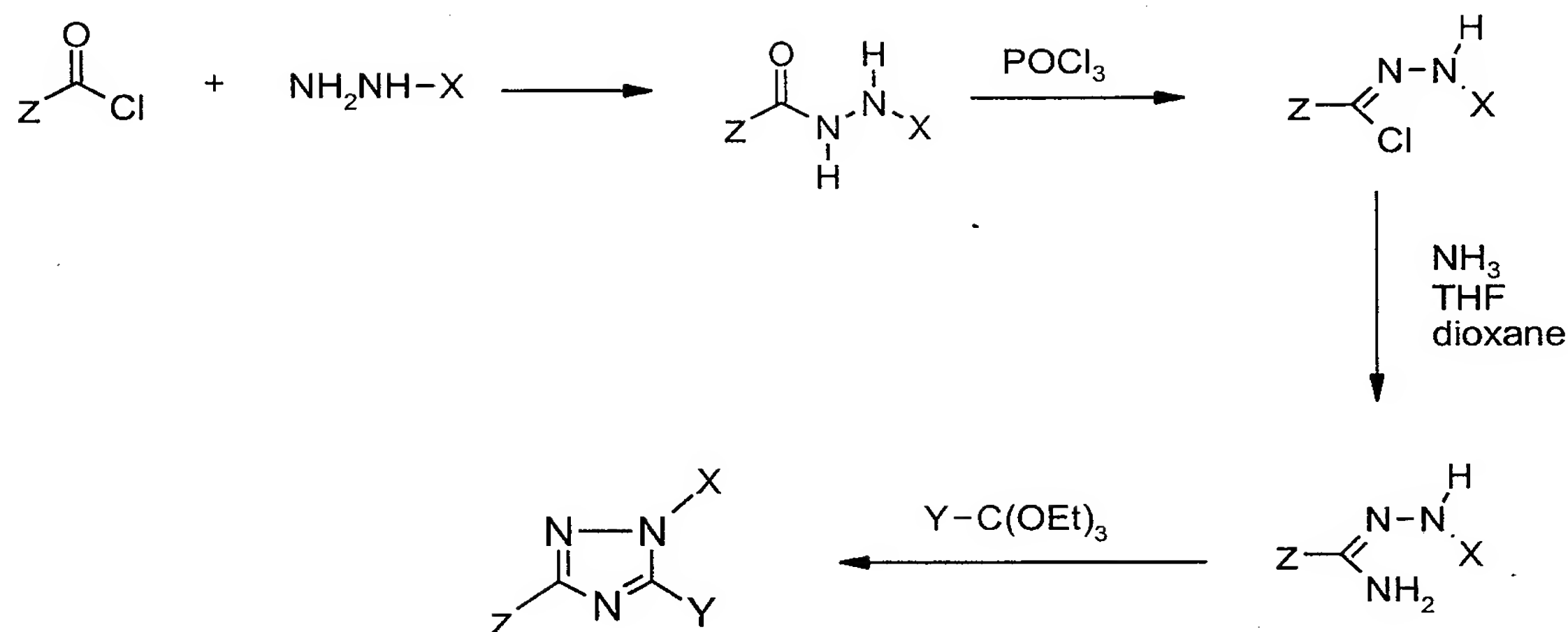
<sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 8.6 (s, 2H), 7.7 (d, J=8.4 Hz, 2H), 7.5  
10 (d, J=8.4 Hz, 2H), 4.1 (s, 3H).

Examples 16-19 illustrate preparation of compounds of formula (1) wherein Y is H, lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, alkoxyalkyl, phenyl, or  
15 substituted phenyl; and X is a group selected from



In general, such compounds are prepared in accordance with the following scheme II:



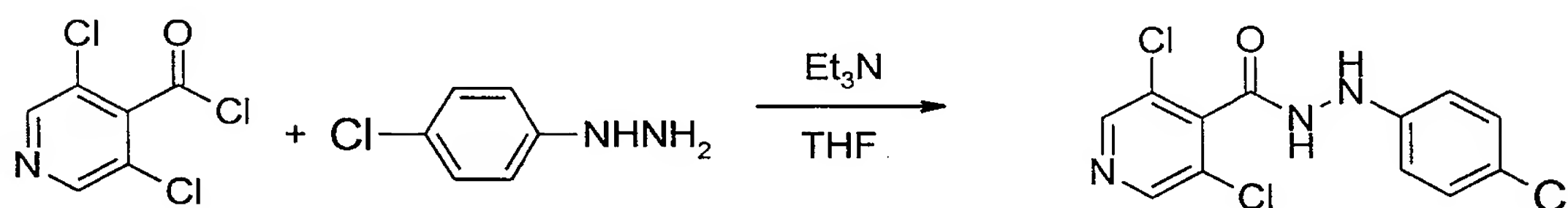


Typical reaction conditions are illustrated in Examples 16-19.

5

## Example 16

3,5-Dichloropyridine-4-carboxylic acid, 4-chlorophenylhydrazide



Triethylamine (1.40 mL, 1.02 g, 10.1 mmol) was added to a suspension of 4-chlorophenylhydrazine hydrochloride (1.83 g, 98%, 10.0 mmol) in THF (75 mL). The mixture was stirred under nitrogen at room temperature for 2.75 h. More triethylamine (3.0 mL, 21.6 mmol) was added followed by dropwise addition of a solution of freshly-prepared 3,5-dichloropyridine-4-carbonyl chloride (10.0 mmol) in THF (20 mL). After stirring overnight, the reaction mixture was concentrated under reduced pressure. The residue was partitioned between ethyl acetate (100 mL) and water (50 mL). The layers were shaken together and allowed to separate. The organic phase was dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated, leaving 3.04 g of a tan solid, mp 201 - 203.5 °C. Recrystallized from ethyl acetate/cyclohexane

and then from ethanol furnished crystals which melted at 206 - 207.5 °C.

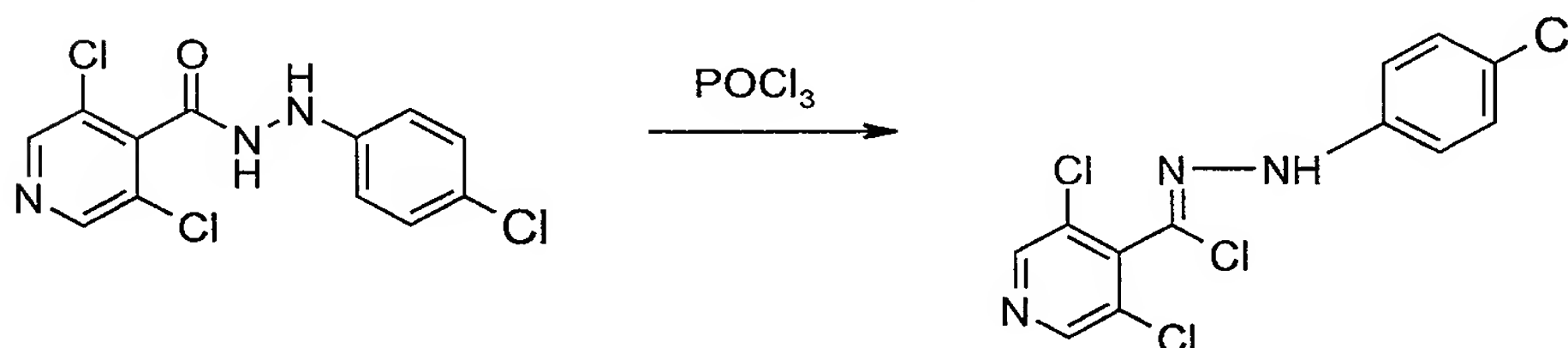
Anal. Calcd for  $C_{12}H_8Cl_3N_3O$ : C, 45.53; H, 2.55; N, 13.27.

Found: C, 45.59; H, 2.49; N, 13.15.

- 5  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  6.80 - 7.24 (m, 4H), 8.40 - 8.56 (m, 2H); IR (nujol)  $\nu_{max}$  3141, 1660  $cm^{-1}$ ; MS: m/e 315 ( $M^+$ ).

#### Example 17

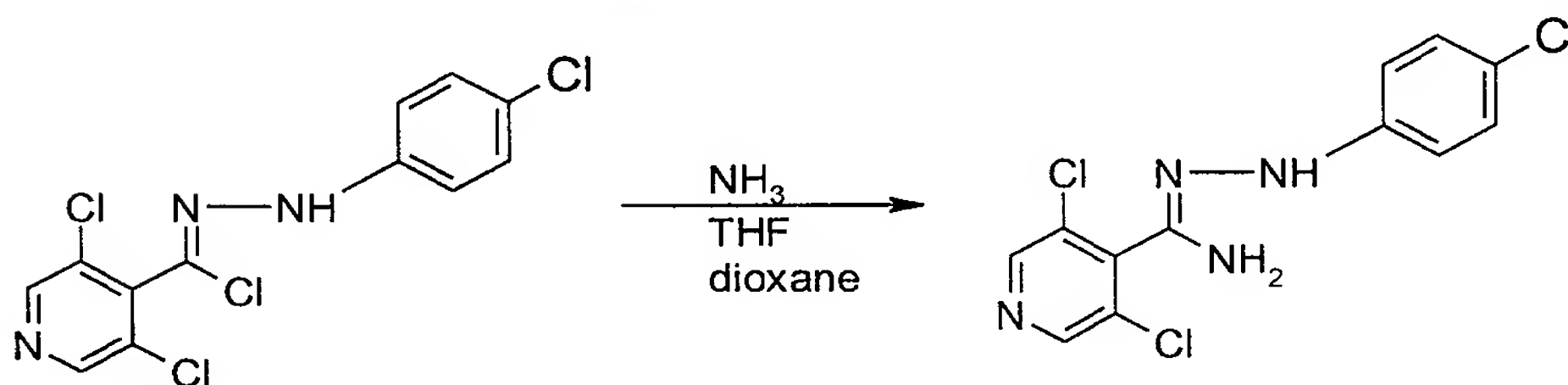
10 3,5-Dichloropyridine-4-carbonyl chloride, 4-chlorophenylhydrazone



A mixture of 3,5-dichloropyridine-4-carboxylic acid, 4-chlorophenylhydrazide (1.50 g, 4.74 mmol) and phosphorus oxychloride (10 mL) containing two drops of DMF was stirred at reflux for 2 h. The clear solution was cooled, concentrated *in vacuo*, and the resulting heavy oil was used without further purification.

#### Example 18

20  $N^1$  - (4-Chlorophenyl) - 3,5-dichloro-4-pyridylcarboxamidrazone

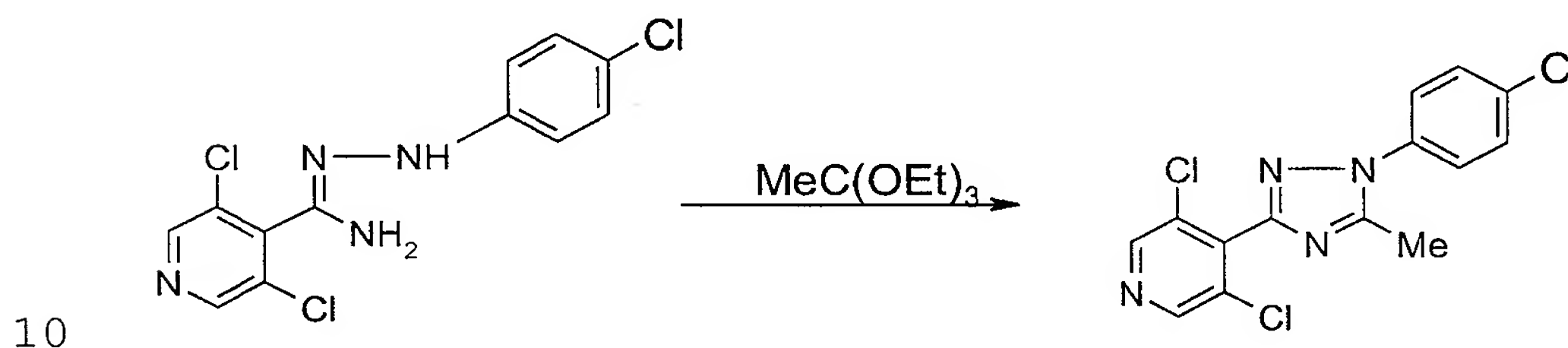


To a solution of 3,5-dichloropyridine-4-carbonyl chloride, 4-chlorophenylhydrazone in THF (40 mL) stirred

under nitrogen at -50 °C was added dropwise a solution of ammonia in dioxane (34.5 mL, 0.5 M, 17.2 mmol). During the addition, the temperature rose to 0 °C. When the addition was complete, the mixture was stirred 15 min and then concentrated under reduced pressure. The residue was used without further purification.

## Example 19

1-(4-Chlorophenyl)-5-methyl-3-(3,5-dichloro-4-pyridyl)-  
1H-1,2,4-triazole



N<sup>1</sup> - (4-Chlorophenyl)-3,5-dichloro-4-pyridyl carboxamidrazone was stirred at reflux in triethyl orthoacetate (15 mL) for 4 h. The reaction mixture was evaporated under reduced pressure and the residue was partitioned between ether and water. The layers were shaken together, separated, and the aqueous phase was extracted with ether. The combined ether layers were dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated under reduced pressure. The residue was triturated with ether and filtered. Concentration of the filtrate left a pasty solid, which was chromatographed on silica, eluting with 20% ether in hexane. The ether percentage was increased to 27% in two increments. Evaporation of appropriate fractions gave 180 mg of the triazole, mp 139 - 144 °C. A sample from a separate lot melted at 142 - 145 °C following recrystallization from cyclohexane.

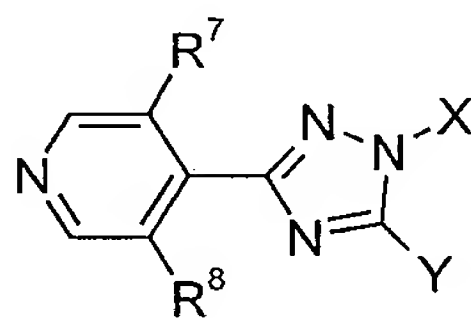
<sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) δ 8.64 (s, 2H), 7.54 (m, 4H), 2.66 (s, 3H); MS: m/e 338 (M<sup>+</sup>).

Anal. Calcd for  $C_{14}H_9Cl_3N_4$ : C, 49.51; H, 2.67; N, 16.50.  
 Found: C, 49.62; H, 2.61; N, 16.39.

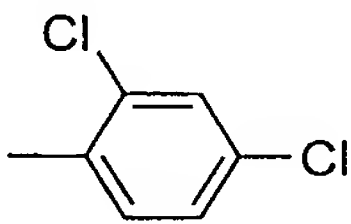
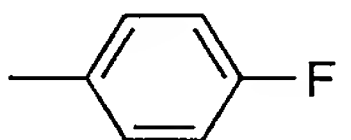
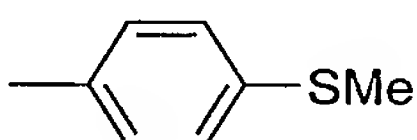
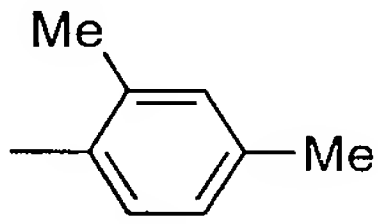
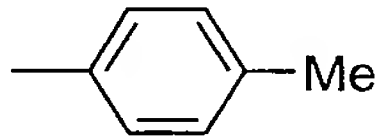
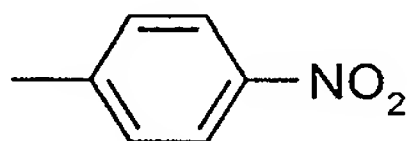
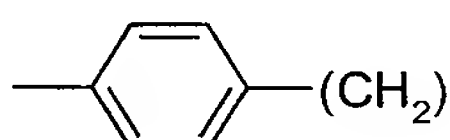
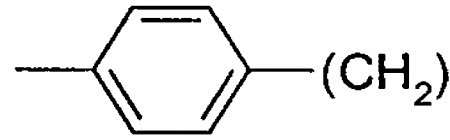
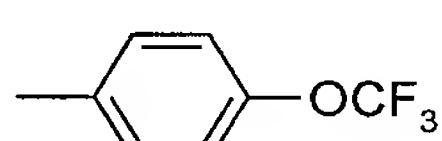
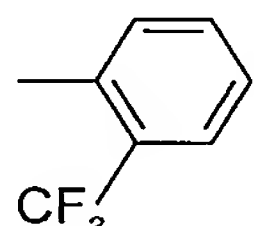
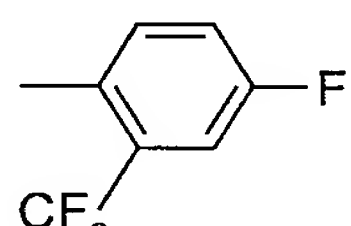
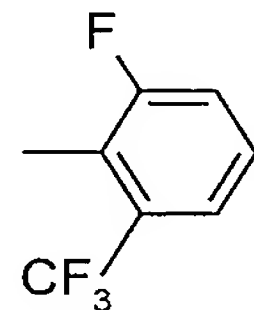
Phytologically acceptable acid addition salts of the compounds of formula (1) are also within the scope of the invention. For example, boron tetrafluoride, hydrogen chloride, hydrogen bromide, hydrogen iodide, hydrogen sulfate, or organic acid salts may be used.

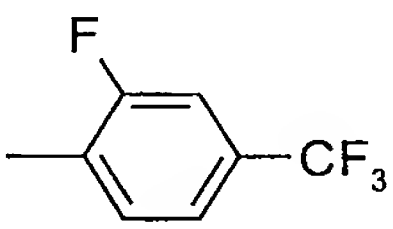
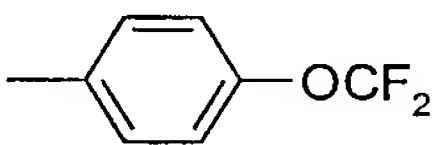
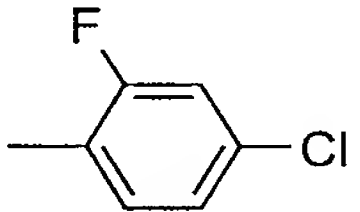
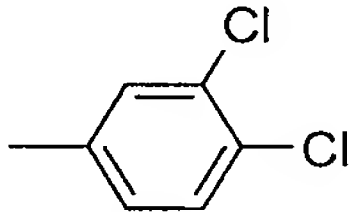
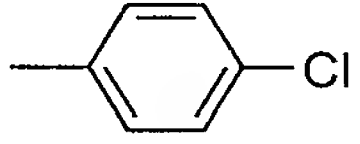
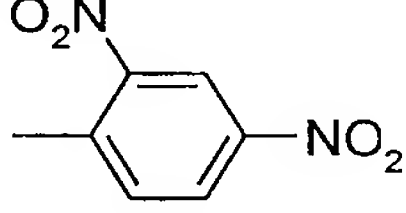
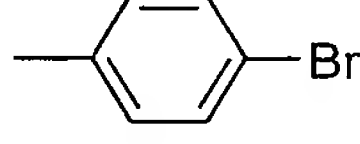
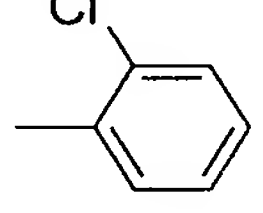
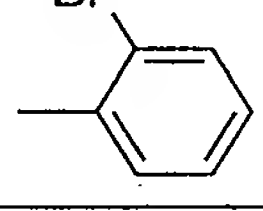
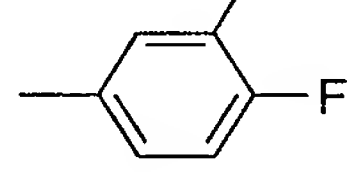
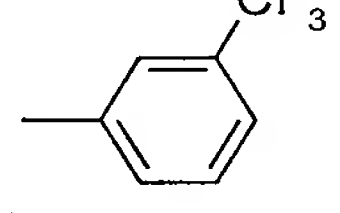
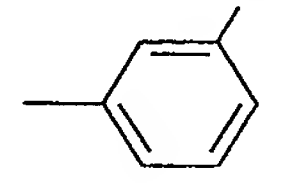
The compounds identified in the following Tables 1-3 were prepared using the procedures illustrated in the foregoing examples, and the compounds were tested against cotton aphid, two-spotted spider mite and sweetpotato whitefly using procedures described hereinafter.

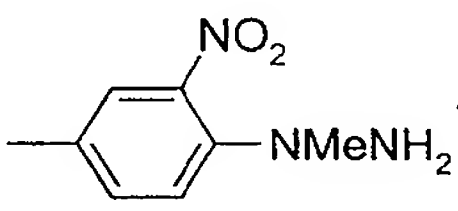
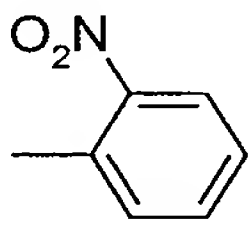

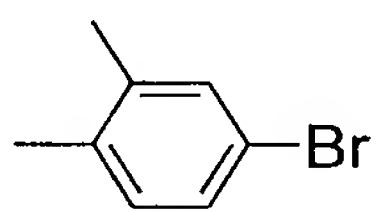
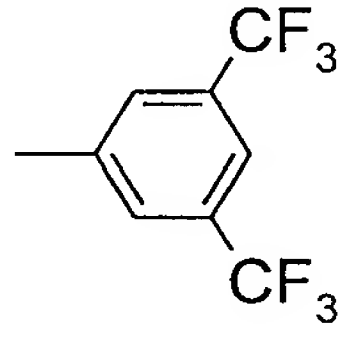
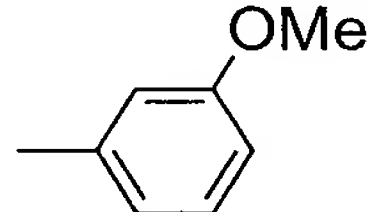
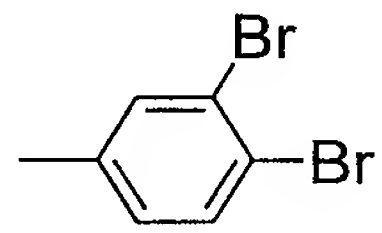
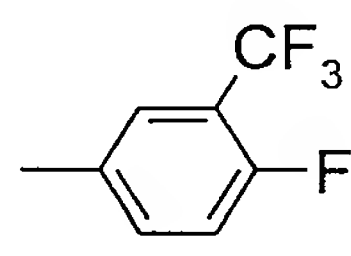
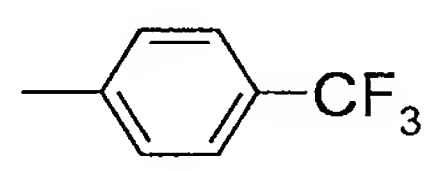
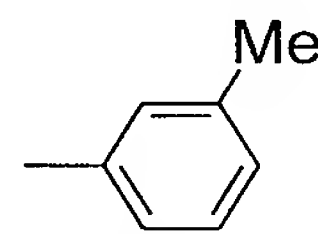
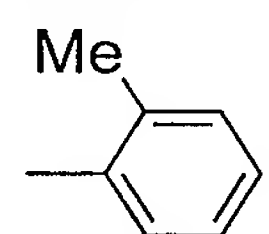
Table 1

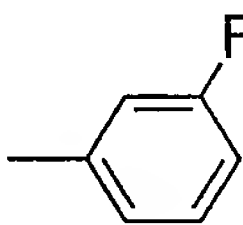
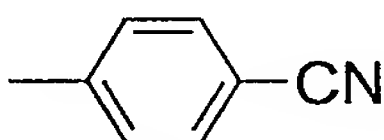
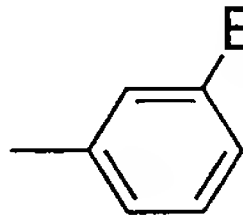
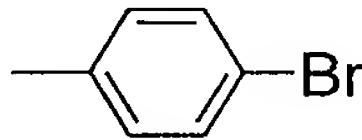
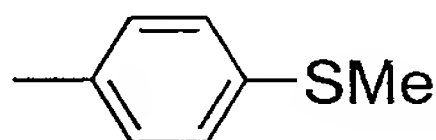
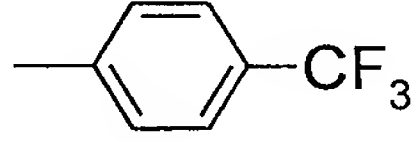

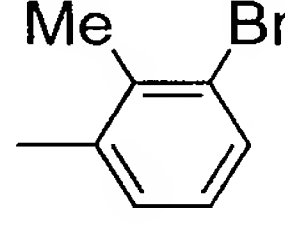
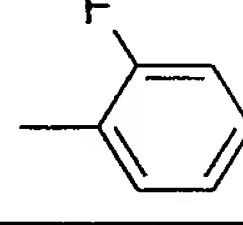
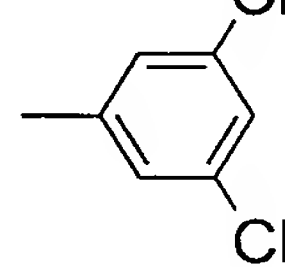
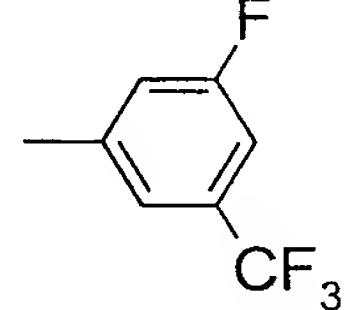



Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
1	Cl	Cl	CH <sub>3</sub>		130-135	A	A	A
2	Cl	Cl	CH <sub>3</sub>		oil	B	A	F
3	Cl	Cl	CH <sub>3</sub>		81-83	B	A	A
4	Cl	Cl	CH <sub>3</sub>		251-252	E	B	F
5	Cl	Cl	CH <sub>3</sub>		oil	C	A	F
6	Cl	Cl	CH <sub>3</sub>		110-112	B	A	A

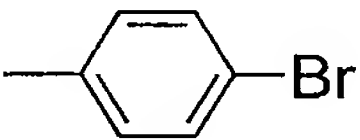
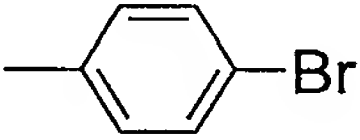
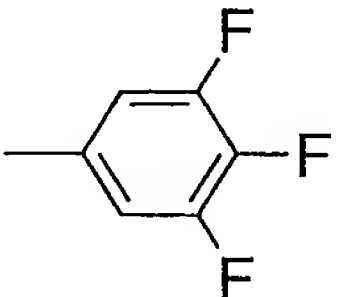
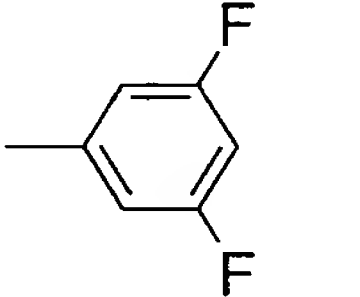
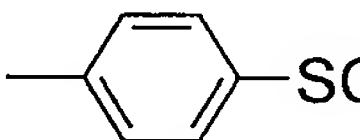
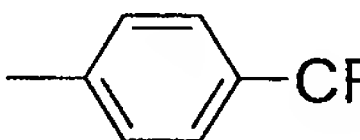
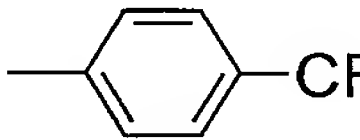
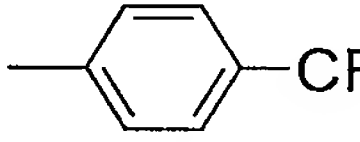
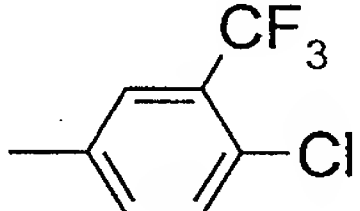
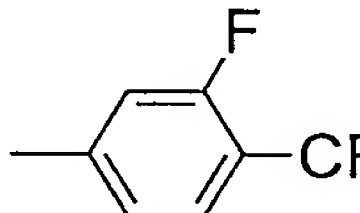
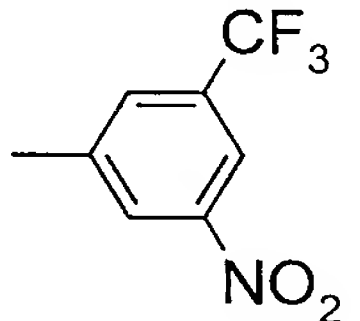
Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
7	Cl	Cl	CH <sub>3</sub>		79-81	B	A	A
8	Cl	Cl	CH <sub>3</sub>		123-126	B	A	C
9	Cl	Cl	CH <sub>3</sub>		oil	B	A	G
10	Cl	Cl	CH <sub>3</sub>		oil	C	A	D
11	Cl	Cl	CH <sub>3</sub>		oil	B	A	D
12	Cl	Cl	CH <sub>3</sub>		162-163	C	A	C
13	Cl	Cl	CH <sub>3</sub>		oil	B	A	F
14	Cl	Cl	CH <sub>3</sub>		oil	C	A	F
15	Cl	Cl	CH <sub>3</sub>		Oil	B	A	A
16	Cl	Cl	CH <sub>3</sub>		Oil	B	A	B
17	Cl	Cl	CH <sub>3</sub>		Oil	E	A	B
18	Cl	Cl	CH <sub>3</sub>		124-126	F	F	G

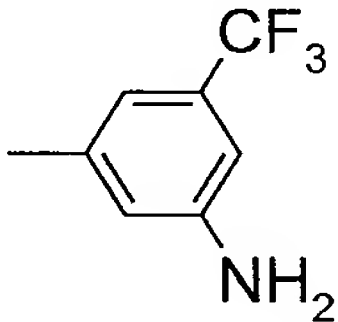
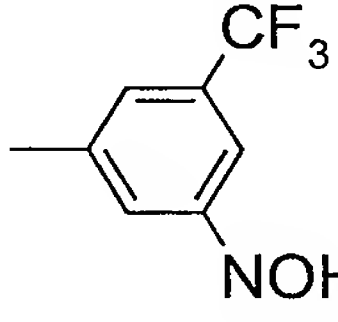

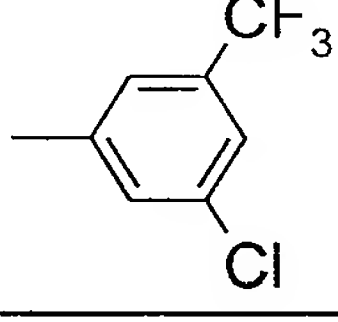
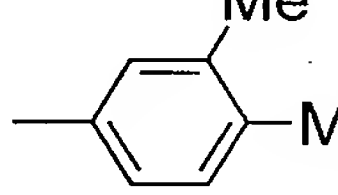
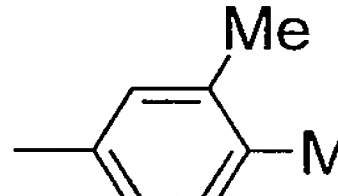
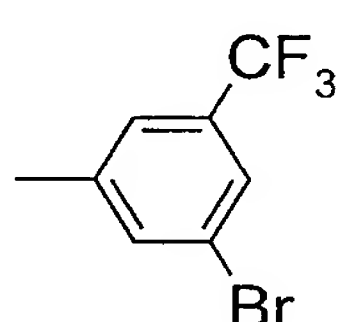
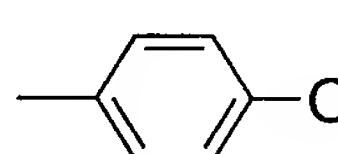

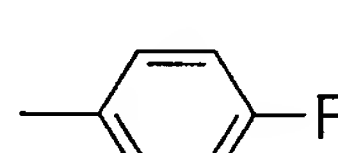
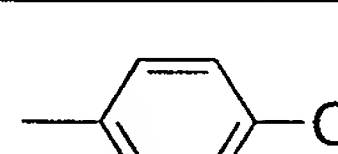
Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
19	Cl	Cl	CH <sub>3</sub>		Oil	B		F
20	Cl	Cl	CH <sub>3</sub>		70-77	A	A	B
21	Cl	Cl	CH <sub>3</sub>		110-116	A	A	C
22	Cl	Cl	CH <sub>3</sub>		Oil	A	A	A
23	Cl	F	CH <sub>3</sub>		Oil	A	A	D
24	Cl	Cl	CH <sub>3</sub>		65-70	B	B	C
25	Cl	OMe	CH <sub>3</sub>		Oil	B		F
26	Cl	Cl	CH <sub>3</sub>		Oil	A	A	A
27	Cl	Cl	CH <sub>3</sub>		Oil	B	A	A
28	Cl	Cl	CH <sub>3</sub>		Oil	A	A	A
29	Cl	Cl	CH <sub>3</sub>		Oil	A	A	F
30	Cl	Cl	CH <sub>3</sub>		Oil	B	A	E


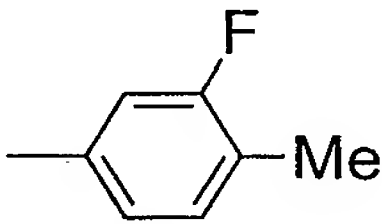
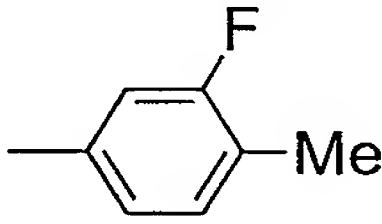
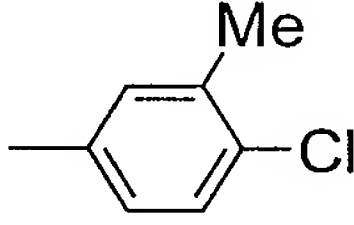
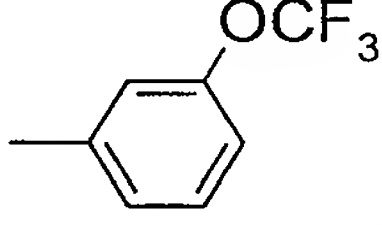
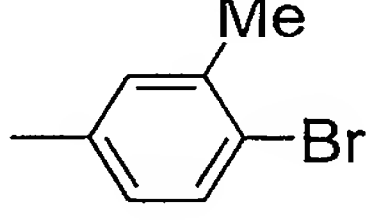
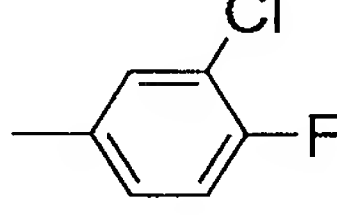
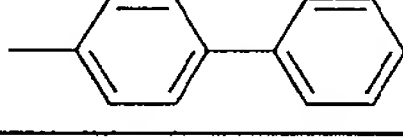
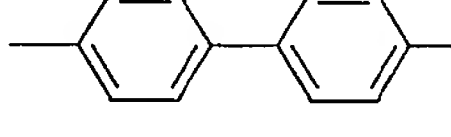
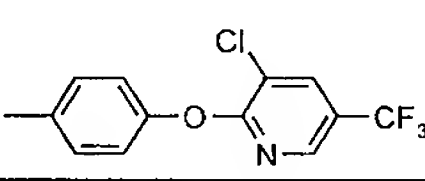
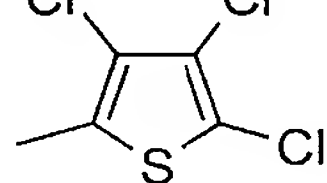
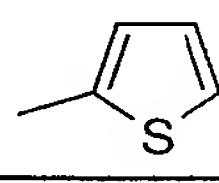
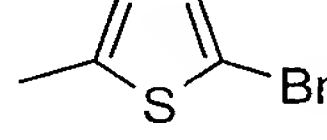
Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
31	Cl	Cl	CH <sub>3</sub>		Oil	F	D	G
32	Cl	Cl	CH <sub>3</sub>		Oil	C	A	F
33	Cl	Cl	CH <sub>3</sub>		Oil	B	A	A
34	Cl	Cl	CH <sub>3</sub>		97-100	A	A	F
35	Cl	Cl	CH <sub>3</sub>		95-97	C	F	F
36	Cl	Cl	CH <sub>3</sub>		Oil	B	A	F
37	Cl	Cl	CH <sub>3</sub>		Oil	A	A	F
38	Cl	Cl	CH <sub>3</sub>		Oil	A	A	F
39	Cl	Cl	H		194-197	G	A	F
40	Cl	Cl	CH <sub>3</sub>		Oil	B	G	F
41	Cl	Cl	CH <sub>3</sub>		Oil	A	G	F

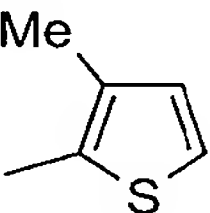
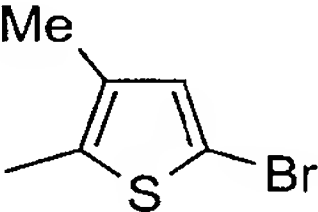
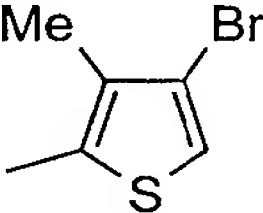
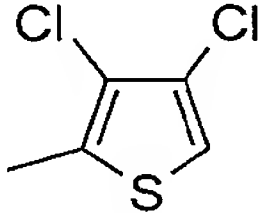
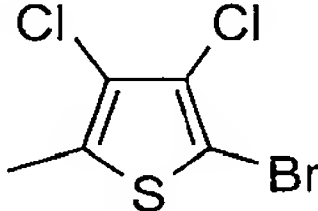
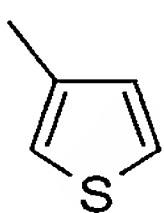
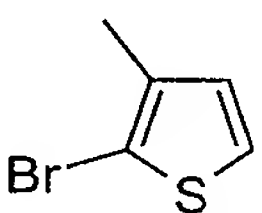

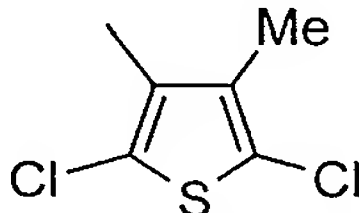
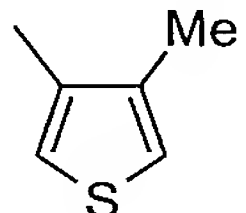
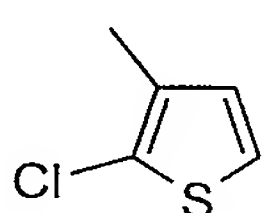
Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
42	Cl	Cl	CH <sub>3</sub>		Oil	A	G	F
43	Cl	Cl	CH <sub>3</sub>		Oil	B	A	E
44	Cl	Cl	CH <sub>3</sub>		Oil	A	A	F
45	Cl	SMe	CH <sub>3</sub>		126-128	B	B	F
46	Cl	SMe	CH <sub>3</sub>		Oil	A		E
47	Cl	SMe	CH <sub>3</sub>		Oil	C	F	E
48	Cl	H	CH <sub>3</sub>		117-120	B	G	C
49	Cl	Cl	CH <sub>3</sub>		67-70	A	A	B
50	Cl	Cl	CH <sub>3</sub>		Oil	B	A	F
51	Cl	Cl	CH <sub>3</sub>		110-113	B	A	C
52	Cl	Cl	CH <sub>3</sub>		Oil	B	A	F
53	Cl	Cl	Et		123-125	B	A	F

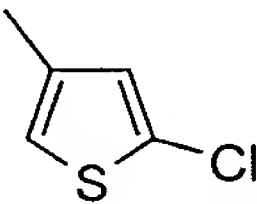
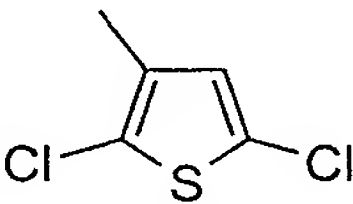
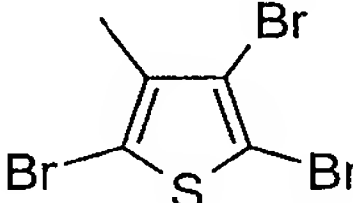
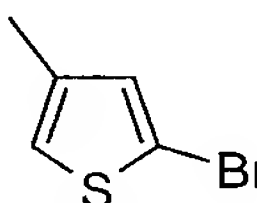
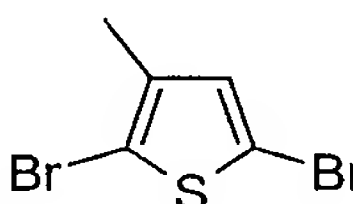
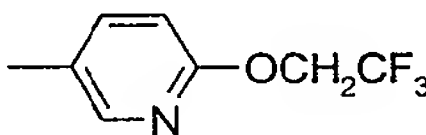
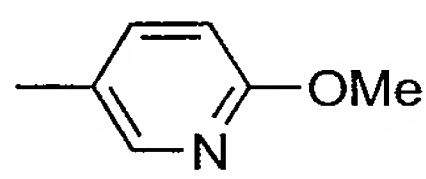
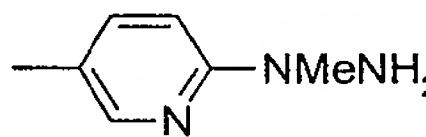
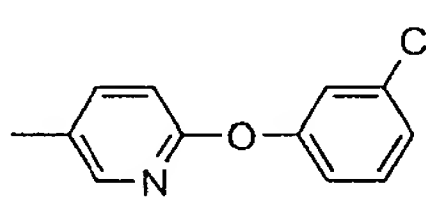
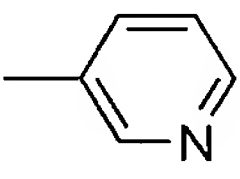
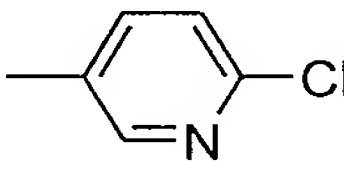
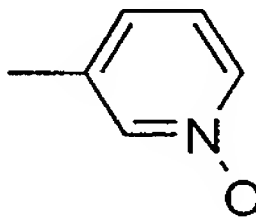


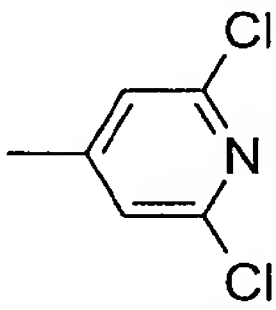
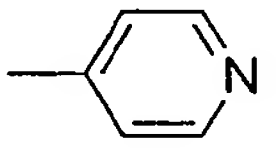
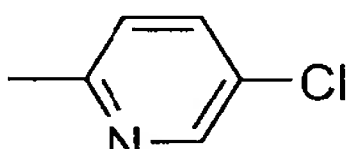
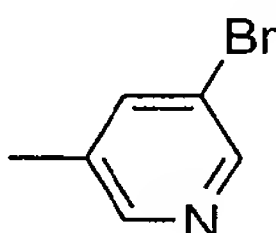
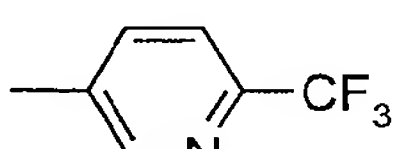
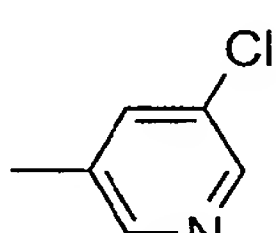
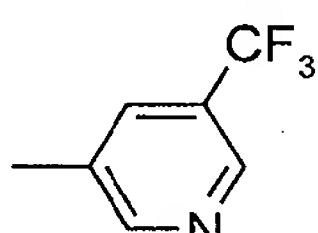
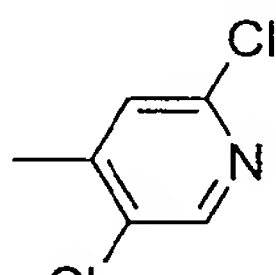
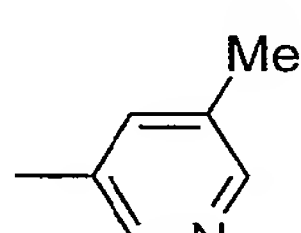
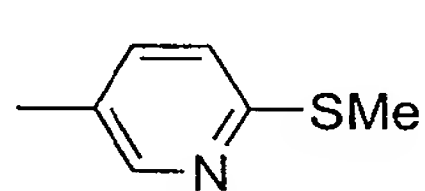
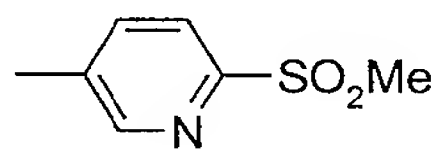
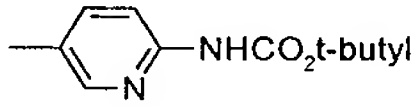
Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
54	Cl	Cl	H		183-185	G	G	F
55	Cl	Cl	n-pentyl		Oil	F	C	F
56	Cl	Cl	CH <sub>3</sub>		Oil	B	A	E
57	Cl	Cl	CH <sub>3</sub>		Oil	B	A	E
58	Cl	Cl	CH <sub>3</sub>		Oil	A	A	D
59	SMe	SMe	CH <sub>3</sub>		134-137	F	F	F
60	Cl	SOMe	CH <sub>3</sub>		182-183	F	A	F
61	Cl	Me	CH <sub>3</sub>		Oil	A	A	F
62	Cl	Cl	CH <sub>3</sub>		Oil	B	A	F
63	Cl	Cl	CH <sub>3</sub>		Oil	A	A	F
64	Cl	Cl	CH <sub>3</sub>		113-118	F	G	F

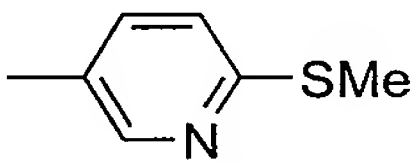
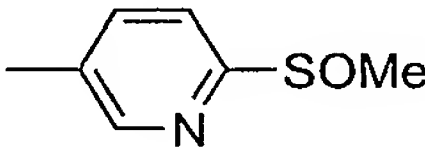
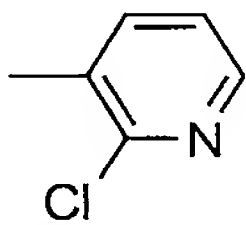
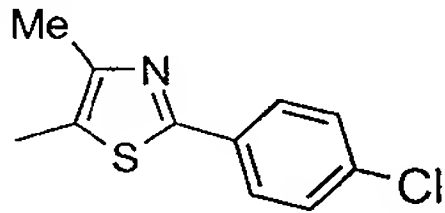
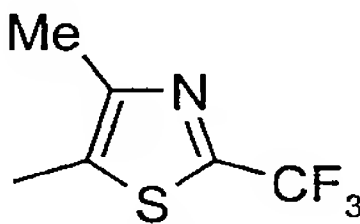
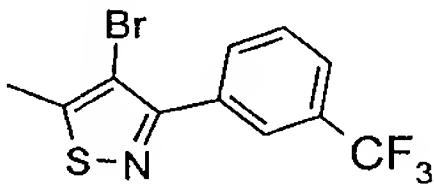
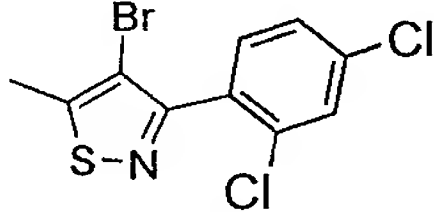
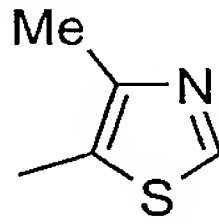
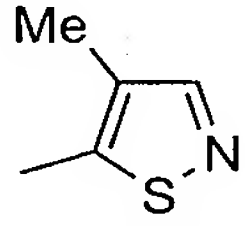
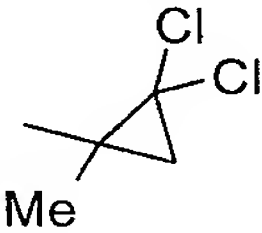
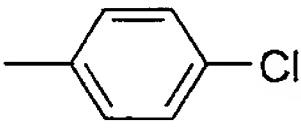
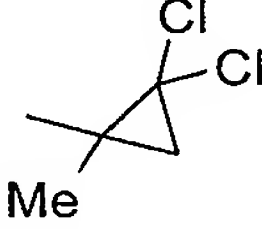
Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA†	TSSM‡	WF*
65	Cl	Cl	CH <sub>3</sub>		201-203	F	G	F
66	Cl	Cl	CH <sub>3</sub>		184-186	F	E	F
67	Cl	Cl	Cyclohexyl		137-139	F	G	F
68	Cl	Cl	CH <sub>3</sub>		103-106	B	F	B
69	Cl	Cl	CH <sub>3</sub>		83-85	E	A	F
70	Cl	Cl	H		236-238	F	F	F
71	Cl	Cl	CH <sub>3</sub>		120-122	A	F	A
72	F	F	CH <sub>3</sub>		82-85	B		E
73	Cl	F	CH <sub>3</sub>		Oil	A	A	F
74	F	F	CH <sub>3</sub>		114-115	B	A	B
75	F	F	CH <sub>3</sub>			A		F

Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
76	Cl	F	CH <sub>3</sub>		74-75	A		D
77	Cl	Cl	H		Oil	G	B	F
78	Cl	Cl	CH <sub>3</sub>		Oil	A	A	F
79	Cl	Cl	CH <sub>3</sub>		111-114	A	A	F
80	Cl	Cl	CH <sub>3</sub>		Oil	A	A	A
81	Cl	Cl	CH <sub>3</sub>		Oil	A	A	F
82	Cl	Cl	CH <sub>3</sub>		Oil	B	A	A
83	Cl	Cl	CH <sub>3</sub>		166-168	A	A	F
84	Cl	Cl	CH <sub>3</sub>		101-104	C	A	G
85	Cl	Cl	CH <sub>3</sub>		oil	B	B	G
86	Cl	Cl	CH <sub>3</sub>		135-137	C	A	F
87	Cl	Cl	CH <sub>3</sub>		oil	C	A	F
88	Cl	Cl	CH <sub>3</sub>		143-148	A	A	B

Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
89	Cl	Cl	CH <sub>3</sub>		oil	E	A	F
90	Cl	Cl	CH <sub>3</sub>		oil	D	A	F
91	Cl	Cl	CH <sub>3</sub>		114-115	C	A	F
92	Cl	Cl	CH <sub>3</sub>		101-102	D	A	F
93	Cl	Cl	CH <sub>3</sub>		Oil	A	A	F
94	Cl	Cl	CH <sub>3</sub>		120-121	F	G	F
95	Cl	Cl	CH <sub>3</sub>		115-120	C	A	F
96	Cl	Cl	CH <sub>3</sub>		oil	C		
97	Cl	Cl	CH <sub>3</sub>		Oil	E		
98	Cl	Cl	CH <sub>3</sub>		100-102	E	A	F
99	Cl	Cl	CH <sub>3</sub>		Oil	B	A	F

Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
100	Cl	Cl	CH <sub>3</sub>		Oil	B		B
101	Cl	Cl	CH <sub>3</sub>		Oil	C	A	F
102	Cl	Cl	CH <sub>3</sub>		Oil	F	A	F
103	Cl	Cl	CH <sub>3</sub>		Oil	A	A	F
104	Cl	Cl	CH <sub>3</sub>		Oil	D	F	F
105	Cl	Cl	CH <sub>3</sub>		oil	B	A	F
106	Cl	Cl	CH <sub>3</sub>		oil	B	A	F
107	Cl	Cl	CH <sub>3</sub>		oil	F	G	G
108	Cl	Cl	CH <sub>3</sub>		oil	C	A	G
109	Cl	Cl	CH <sub>3</sub>		106-116	D	A	G
110	Cl	Cl	CH <sub>3</sub>			A	A	D
111	Cl	Cl	CH <sub>3</sub>			D	E	F

Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
112	Cl	Cl	CH <sub>3</sub>		162-164	B	E	E
113	Cl	Cl	CH <sub>3</sub>		130-132	C	A	F
114	Cl	Cl	CH <sub>3</sub>		106-107	A	A	F
115	Cl	Cl	CH <sub>3</sub>		140-144	A	A	F
116	Cl	Cl	CH <sub>3</sub>		117-120	A	A	F
117	Cl	Cl	CH <sub>3</sub>		Oil	A	A	E
118	Cl	Cl	CH <sub>3</sub>		Oil	B	A	F
119	Cl	Cl	CH <sub>3</sub>		129-135	B	A	F
120	Cl	Cl	CH <sub>3</sub>		126-128	F	A	F
121	Cl	Cl	CH <sub>3</sub>		121-123		B	F
122	Cl	Cl	CH <sub>3</sub>		183-186	F	G	G
123	Cl	Cl	CH <sub>3</sub>		Oil	G	A	F

Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
124	Cl	Cl	H		198-200	G		F
125	Cl	Cl	CH <sub>3</sub>		Oil	F		E
126	Cl	Cl	CH <sub>3</sub>		Oil	B	A	B
127	Cl	Cl	CH <sub>3</sub>		185-187	F	A	F
128	Cl	Cl	CH <sub>3</sub>		136-138	A	A	F
129	Cl	Cl	CH <sub>3</sub>		56-59	A	A	F
130	Cl	Cl	CH <sub>3</sub>		127-129	B	D	F
131	Cl	Cl	CH <sub>3</sub>		143-145	B	A	F
132	Cl	Cl	CH <sub>3</sub>		82-84	F	F	F
133	Cl	Cl	CH <sub>3</sub>		79-83	E	A	A
134	Cl	Cl			117-119	G	A	F

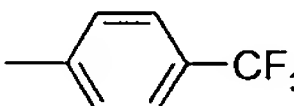
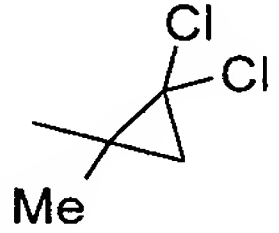
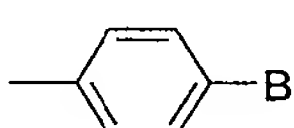
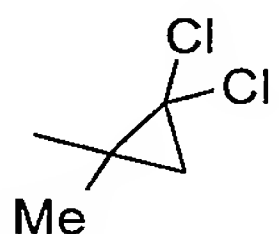
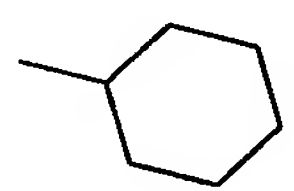
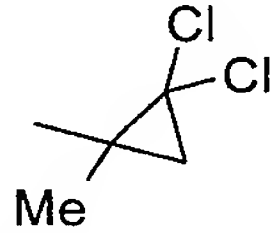
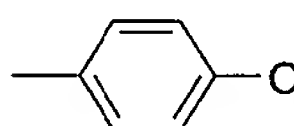
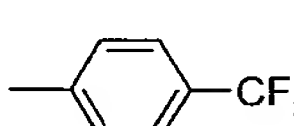
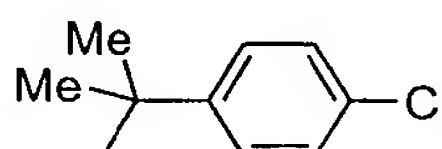
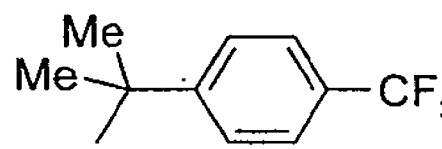
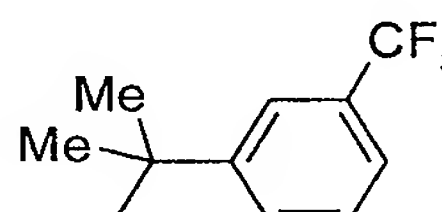
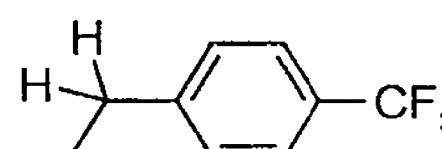
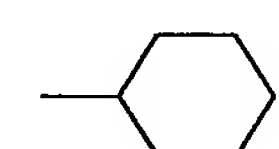
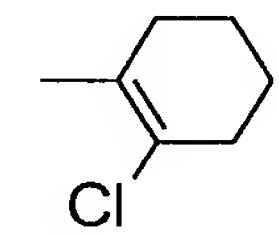
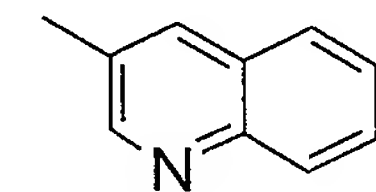
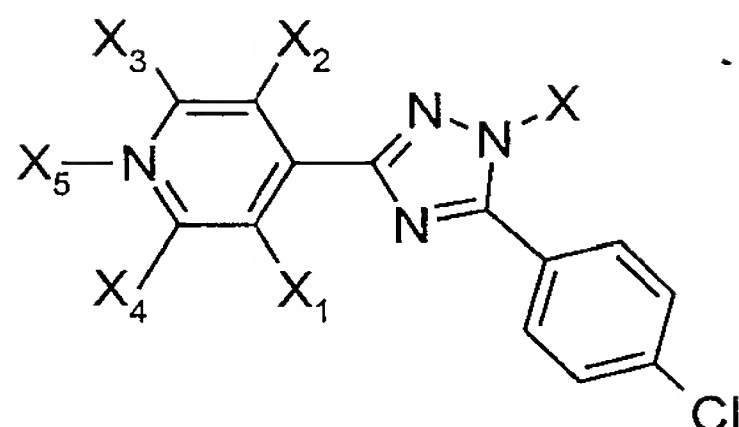
Cmpd No.	R <sup>7</sup>	R <sup>8</sup>	X	Y	mp °C	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
135	Cl	Cl			158-160	F	A	F
136	Cl	Cl			138-148	G	B	D
137	Cl	Cl			139-142	F	G	F
138	Cl	Cl		Me	144-146	B	A	A
139	Cl	Cl		Me	Oil	C	A	A
140	Cl	Cl	CH <sub>3</sub>		oil	F	A	F
141	Cl	Cl	CH <sub>3</sub>		96-101	D	A	E
142	Cl	Cl	CH <sub>3</sub>		98 ½-100	F	A	F
143	Cl	Cl	CH <sub>3</sub>		Oil	B		F
144	Cl	Cl	CH <sub>3</sub>		Oil	C	A	F
145	Cl	Cl	CH <sub>3</sub>		117-121	E	A	D
146	Cl	Cl	CH <sub>3</sub>		160-164	B	A	E

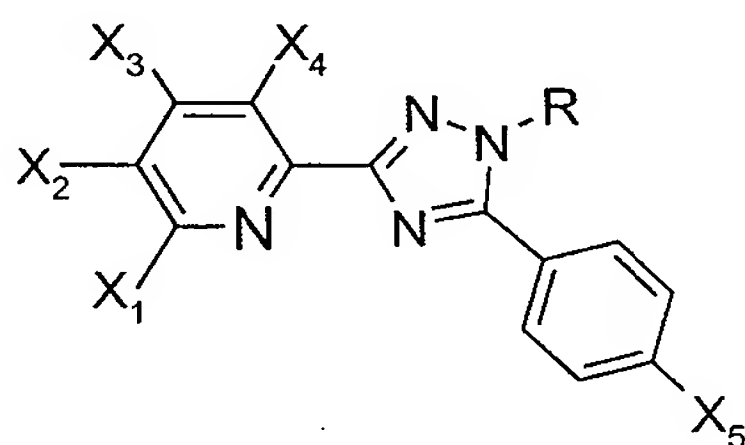


Table 2



cmpd no.	X1	X2	X3	X4	X5	X	mp	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
148	Cl	Cl	Cl	Cl		Me	129-134	F		
149	Cl	Cl	H	H	O	Me	161-166	A	A	D
150	F	F	F	F		Me	oil	F	G	F
151	Cl	Cl	H	H	O	Me	151-153	A	A	A
152	Cl	Cl	H	H	O	Me	169-173			B
153	Cl	H	Cl	H		Me	133-134	B	G	F
154	Cl	H	F	F		Me	Oil	D	G	D

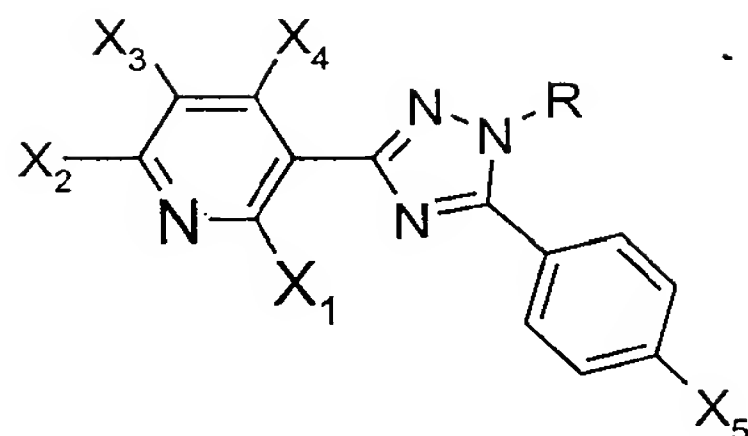
Table 3



5

cmpd no.	X1	X2	X3	X4	X5	R	mp	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
155	F	F	F	F	Cl	Me	154-156	G		
156	Cl	H	H	Cl	Cl	Me	98-102	G	F	
157	H	H	H	Cl	Cl	Me	oil	B	F	G
158	H	Cl	H	Cl	Cl	Me	105-107	F	G	F
159	H	H	H	Cl	Me	Me	113-117	F		F
160	H	H	H	Cl	CF <sub>3</sub>	Me	oil	B	A	G

Table 4



cmpd no.	X1	X2	X3	X4	X5	R	mp	CA <sup>†</sup>	TSSM <sup>‡</sup>	WF <sup>*</sup>
161	Cl	H	H	H	Cl	Me	176-177	F		
162	Cl	H	H	Cl	Cl	Me	oil	A	G	
163	Cl	Cl	H	CF3	Cl	Me	124-126	G	G	F

CA<sup>†</sup> refers to activity at 50 ppm against cotton aphid,

5 TSSM<sup>‡</sup> refers to activity at 100 ppm against two-spotted spider mite, and

In each case the rating scale is as follows

% Control	Rating
91-100	A
81-90	B
71-80	C
61-70	D
51-60	E
less than 51	F
inactive	G

#### Insecticide and Miticide Utility

The compounds of the formulae (1) are suitable for  
 10 controlling pests on animals and plants. Such pests  
 belong mainly to the arthropod family, such as,  
 especially, insects of the orders Lepidoptera,  
 Coleoptera, Homoptera, Heteroptera, Diptera,  
 Thysanoptera, Orthoptera, Anoplura, Siphonaptera,  
 15 Mallophaga, Thysanura, Isoptera, Psocoptera or

Hymenoptera, and arachnids of the order Acarina, such as, for example, mites, aphids, and ticks.

Therefore, the present invention also is directed to a method for inhibiting an insect, mite, or aphid which comprises applying to a locus of the insect or mite an insect- or mite-inhibiting amount of a compound of formula (1).

The compounds are useful for reducing populations of insects and mites and are useful in a method of inhibiting an insect or mite population which comprises applying to a locus of the insect or mite an effective insect- or mite-inactivating amount of a compound of formula (1). The "locus" of insects or mites is a term used herein to refer to the environment in which the insects or mites live or where their eggs are present, including the air surrounding them, the food they eat, or objects which they contact. For example, plant-ingesting insects or mites can be controlled by applying the active compound to plant parts that the insects or mites eat, particularly the foliage. It is contemplated that the compounds might also be useful to protect textiles, paper, stored grain, or seeds by applying an active compound to such substance. The term "inhibiting an insect or mite" refers to a decrease in the numbers of living insects or mites, or a decrease in the number of viable insect or mite eggs. The extent of reduction accomplished by a compound depends, of course, upon the application rate of the compound, the particular compound used, and the target insect or mite species. At least an inactivating amount should be used. The terms "insect-inactivating amount" and "mite-inactivating amount" are used to describe the amount, which is sufficient to cause a measurable reduction in the treated insect or mite, population. Generally an amount in the range from about 1 to about 1000 ppm active compound is used.

In a preferred embodiment, the present invention is directed to a method for inhibiting a mite or aphid which comprises applying to a plant an effective mite- or aphid- inactivating amount of a compound of formula (1).

5 Insecticidal test for cotton aphid (*Aphis gossypii*):

To prepare spray solutions, 1 mg of each test compound was dissolved into 1 mL of a 90:10 acetone:ethanol solvent. This 1 mL of chemical solution was added to 19 mL of water containing 0.05% Tween 20  
10 surfactant to produce a 50 ppm spray solution.

Squash cotyledons were infested with cotton aphid (all life stages) 16-20 hours prior to application of spray solution. The solution was sprayed on both sides of each infested squash cotyledon (0.5 mL X 2 each side)  
15 with a sweeping action until runoff. The plants were allowed to air dry and held for 3 days in a controlled room at 26° C and 40 % relative humidity after which time the test was graded. Grading was by actual count using a dissecting microscope and comparison of test counts to  
20 the untreated check. Results are given in Tables 1-3 as percent control based on population reduction versus the untreated.

Insecticidal test for two-spotted spider mite  
(*Tetranychus urticae*)

25 Ovicide Method:

Ten adult female two-spotted spider mites were placed on eight 2.2 cm leaf discs of cotton leaf, allowed to oviposit over 24 hours, and thereafter removed. The leaf discs were sprayed with 100 ppm test solutions using a  
30 hand syringe, then allowed to dry with sixteen discs left untreated as a negative control. Discs were placed on an agar substrate and held at 24 °C and 90% relative humidity for 6 days. Percent control based on the number of

hatched larvae on treated discs and the number on untreated discs is reported in Tables 1-2.

Evaluation of Tests Compounds on Sweetpotato Whitefly  
(*Bemisia tabacia*) Under Laboratory Conditions

5           Four mg of each test compound was dissolved by  
adding 4 ml of a 90:10 acetone:ethanol solvent mixture to  
the vial containing the sample compound. This solution  
was added to 16 ml of water containing 0.05% Tween 20  
surfactant to produce  
10   20 ml of an 200 ppm spray solution.

Five-week-old cotton plants reared in a greenhouse  
were stripped of all foliage except for the two uppermost  
true leaves that were greater than 5 cm in diameter.  
These plants were then placed into a laboratory colony of  
15   whiteflies for two days for oviposition by the colony  
females. All whiteflies were then removed from the test  
plants with pressurized air. The spray solution was then  
applied to the test plants with a hand-held syringe  
fitted with hollow cone nozzle. One mL spray solution  
20   was applied to each leaf top and bottom for a total of 4  
mL per plant. Four replications of each test compound  
utilized a total of 16 mL spray solution. Plants were  
air dried and then placed in a holding chamber (28° C and  
60% RH) for 13 days. Compound efficacy was evaluated by  
25   counting, under an illuminated magnifying glass, the  
number of large nymphs (3rd-4th instar) per leaf.

Percent control based on reduction of large nymphs of a  
test compound compared to solution-only (no test  
compound) sprayed plants is reported in Tables 1-3.

30           Compounds of the invention have demonstrated  
unexpectedly good activity against citrus red mite, and  
have also demonstrated unique female sterilization

activity against mites, when tested in the following methods:

**Citrus Red Mite Ovicide:**

Ten adult female citrus red mites *Panonychus citri* were placed on six 2cm leaf discs of orange leaf maintained on an agar substrate. They were allowed to oviposit over 24 hours and then removed by aspiration. The leaf discs were sprayed with test solutions using a hand syringe using a TN-3 nozzle just to wetting, then allowed to dry. Twelve discs were treated with water as a negative control. Discs were maintained on an agar substrate and held at 27 degrees C and 90% relative humidity for 6 days. Ovicidal activity was calculated by comparing the number of hatched larvae on treated discs with the number on water only treated discs.

**Two-spotted spider mite female Sterilization:**

Adult female two-spotted spider mites *Tetranychus urticae* were sprayed to runoff using the appropriate test solution and allowed to dry. Ten treated females were placed on each of twenty five untreated 2 cm leaf discs of kidney bean. These mites were allowed to oviposit for the selected time interval and then were transferred to new untreated leaf discs. This transfer usually occurred at 24, 48, 72, and 144 hours. Twenty five discs infested with untreated female mites was maintained as a negative control. Discs were held at 27 degrees C and 90% relative humidity for 6 days. Residual tests were assayed over time by infesting the treated plants with female mites over several time intervals. Female mites were exposed to the treated plants for 16 hours and then transferred to untreated discs as above. Ovicidal activity was calculated by comparing the number of hatched larvae from eggs laid by treated females versus untreated females.

Compounds of the invention have shown activity against Lygus, milkweed bug, plant hoppers, and tobacco bud worm.

In addition to being effective against mites, aphids, and insects when applied to foliage, compounds of formula (1) have systemic activity. Accordingly, another aspect of the invention is a method of protecting a plant from insects which comprises treating plant seed prior to planting it, treating soil where plant seed is to be planted, or treating soil at the roots of a plant after it is planted, with an effective amount of a compound of formula (1).

#### Compositions

The compounds of this invention are applied in the form of compositions which are important embodiments of the invention, and which comprise a compound of this invention and a phytologically-acceptable inert carrier. The compositions are either concentrated formulations which are dispersed in water for application, or are dust or granular formulations which are applied without further treatment. The compositions are prepared according to procedures and formulae which are conventional in the agricultural chemical art, but which are novel and important because of the presence therein of the compounds of this invention. Some description of the formulation of the compositions will be given, however, to assure that agricultural chemists can readily prepare any desired composition.

The dispersions in which the compounds are applied are most often aqueous suspensions or emulsions prepared from concentrated formulations of the compounds. Such water-soluble, water-suspendable or emulsifiable formulations are either solids, usually known as wettable powders, or liquids usually known as emulsifiable concentrates or aqueous suspensions. Wettable powders,



which may be compacted to form water dispersible granules, comprise an intimate mixture of the active compound, an inert carrier, and surfactants. The concentration of the active compound is usually from about 10% to about 90% by weight. The inert carrier is usually chosen from among the attapulgite clays, the montmorillonite clays, the diatomaceous earths, or the purified silicates. Effective surfactants, comprising from about 0.5% to about 10% of the wettable powder, are found among the sulfonated lignins, the condensed naphthalenesulfonates, the naphthalenesulfonates, the alkylbenzenesulfonates, the alkyl sulfates, and nonionic surfactants such as ethylene oxide adducts of alkyl phenols.

Emulsifiable concentrates of the compounds comprise a convenient concentration of a compound, such as from about 50 to about 500 grams per liter of liquid, equivalent to about 10% to about 50%, dissolved in an inert carrier which is either a water miscible solvent or a mixture of water-immiscible organic solvent and emulsifiers. Useful organic solvents include aromatics, especially the xylenes, and the petroleum fractions, especially the high-boiling naphthalenic and olefinic portions of petroleum such as heavy aromatic naphtha. Other organic solvents may also be used, such as the terpenic solvents including rosin derivatives, aliphatic ketones such as cyclohexanone, and complex alcohols such as 2-ethoxyethanol. Suitable emulsifiers for emulsifiable concentrates are chosen from conventional nonionic surfactants, such as those discussed above.

Aqueous suspensions comprise suspensions of water-insoluble compounds of this invention, dispersed in an aqueous vehicle at a concentration in the range from about 5% to about 50% by weight. Suspensions are prepared by finely grinding the compound, and vigorously



mixing it into a vehicle comprised of water and surfactants chosen from the same types discussed above. Inert ingredients, such as inorganic salts and synthetic or natural gums, may also be added, to increase the density and viscosity of the aqueous vehicle. It is often most effective to grind and mix the compound at the same time by preparing the aqueous mixture, and homogenizing it in an implement such as a sand mill, ball mill, or piston-type homogenizer.

The compounds may also be applied as granular compositions, which are particularly useful for applications to the soil. Granular compositions usually contain from about 0.5% to about 10% by weight of the compound, dispersed in an inert carrier which consists entirely or in large part of clay or a similar inexpensive substance. Such compositions are usually prepared by dissolving the compound in a suitable solvent and applying it to a granular carrier which has been pre-formed to the appropriate particle size, in the range of from about 0.5 to 3 mm. Such compositions may also be formulated by making a dough or paste of the carrier and compound and crushing and drying to obtain the desired granular particle size.

Dusts containing the compounds are prepared simply by intimately mixing the compound in powdered form with a suitable dusty agricultural carrier, such as kaolin clay, ground volcanic rock, and the like. Dusts can suitably contain from about 1% to about 10% of the compound.

It is equally practical, when desirable for any reason, to apply the compound in the form of a solution in an appropriate organic solvent, usually a bland petroleum oil, such as the spray oils, which are widely used in agricultural chemistry.

Insecticides and acaricides are generally applied in the form of a dispersion of the active ingredient in a

liquid carrier. It is conventional to refer to application rates in terms of the concentration of active ingredient in the carrier. The most widely used carrier is water.

5       The compounds of the invention can also be applied in the form of an aerosol composition. In such compositions the active compound is dissolved or dispersed in an inert carrier, which is a pressure-generating propellant mixture. The aerosol composition  
10 is packaged in a container from which the mixture is dispensed through an atomizing valve. Propellant mixtures comprise either low-boiling halocarbons, which may be mixed with organic solvents, or aqueous suspensions pressurized with inert gases or gaseous  
15 hydrocarbons.

      The actual amount of compound to be applied to loci of insects, mites, and aphids is not critical and can readily be determined by those skilled in the art in view of the examples above. In general, concentrations of  
20 from-10 ppm to 5000 ppm of compound are expected to provide good control. With many of the compounds, concentrations of from 100 to 1500 ppm will suffice. For field crops, such as soybeans and cotton, a suitable application rate for the compounds is about 0.5 to 1.5  
25 lb/A, typically applied in 5-20 gal/A of spray formulation containing 1200 to 3600 ppm of compound. For citrus crops, a suitable application rate is from about 100 to 1500 gal/A spray formulation, which is a rate of 100 to 1000 ppm.

30       The locus to which a compound is applied can be any locus inhabited by an insect or arachnid, for example, vegetable crops, fruit and nut trees, grape vines, and ornamental plants. Inasmuch as many mite species are specific to a particular host, the foregoing list of mite  
35 species provides exemplification of the wide range of

settings in which the present compounds can be used.

Because of the unique ability of mite eggs to resist toxicant action, repeated applications may be desirable to control newly emerged larvae, as is true of other known acaricides.

The following formulations of compounds of the invention are typical of compositions useful in the practice of the present invention.

A. 0.75 Emulsifiable Concentrate

Compound of formula (1)	9.38%
"TOXIMUL D" (nonionic/anionic surfactant blend)	2.50%
"TOXIMUL H" (nonionic/anionic surfactant blend)	2.50%
"EXXON 200" (naphthalenic solvent)	85.62%

B. 1.5 Emulsifiable Concentrate

Compound of formula (1)	18.50%
"TOXIMUL D"	2.50%
"TOXIMUL H"	2.50%
"EXXON 200"	76.50%

C. 1.0 Emulsifiable Concentrate

Compound of formula (1)	12.5%
N-methylpyrrolidone	25.00%
"TOXIMUL D"	2.50%
"TOXIMUL H"	2.50%
"EXXON 200"	57.50%

D. 1.0 Aqueous Suspension

Compound of formula (1)	12.00%
"PLURONIC P-103" (block copolymer of propylene oxide and ethylene oxide, surfactant)	1.50%
"PROXEL GXL" (biocide/preservative)	.05%
"AF-100" (silicon based antifoam agent)	.20%
"REAX 88B" (lignosulfonate dispersing agent)	1.00%
propylene glycol	10.00%
veegum	.75%
xanthan	.25%

water	74.25%
<u>E. 1.0 Aqueous Suspension</u>	
Compound of formula (1)	12.50%
"MAKON 10" (10 moles ethyleneoxide nonylphenol surfactant)	1.00%
"ZEOSYL 200" (silica)	1.00%
"AF-100"	0.20%
"AGRIWET FR" (surfactant)	3.00%
2% xanthan hydrate	10.00%
water	72.30%
<u>F. 1.0 Aqueous Suspension</u>	
Compound of formula (1)	12.50%
"MAKON 10"	1.50%
"ZEOSYL 200" (silica)	1.00%
"AF-100"	0.20%
"POLYFON H" (lignosulfonate dispersing agent)	0.20%
2% xanthan hydrate	10.00%
water	74.60%
<u>G. Wettable Powder</u>	
Compound of formula (1)	25.80%
"POLYFON H"	3.50%
"SELLOGEN HR"	5.00%
"STEPANOL ME DRY"	1.00%
gum arabic	0.50%
"HISIL 233"	2.50%
Barden clay	61.70%
<u>H. 1.0 Aqueous Suspension</u>	
Compound of formula (1)	12.40%
"TERGITOL 158-7"	5.00%
" ZEOSYL 200 "	1.0%
"AF-1G0"	0.20%
"POLYFON H"	0.50%
2% xanthan solution	10.00%
tap water	70.90%

I. 1.0 Emulsifiable Concentrate

Compound of formula (1)	12.40%
"TOXIMUL D"	2.50%
"TOXIMUL H"	2.50%
"EXXON 200"	82.60%

J. Wettable Powder

Compound of formula (1)	25.80%
"SELLOGEN HR"	5.00%
"POLYFON H"	4.00%
"STEPANOL ME DRY"	2.00%
"HISIL 233"	3.00%
Barden clay	60.20%

K. 0.5 Emulsifiable Concentrate

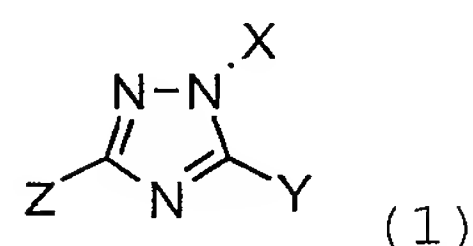
Compound of formula (1)	6.19%
"TOXIMUL H"	3.60%
"TOXIMUL D"	0.40%
"EXXON 200"	89.81%

L. Emulsifiable Concentrate

Compound of formula (1)	5 to 48%
surfactant or surfactant blend	2 to 20%
Aromatic Solvent or Mixture	55 to 75%

# Claims

1. A compound of the formula (1)

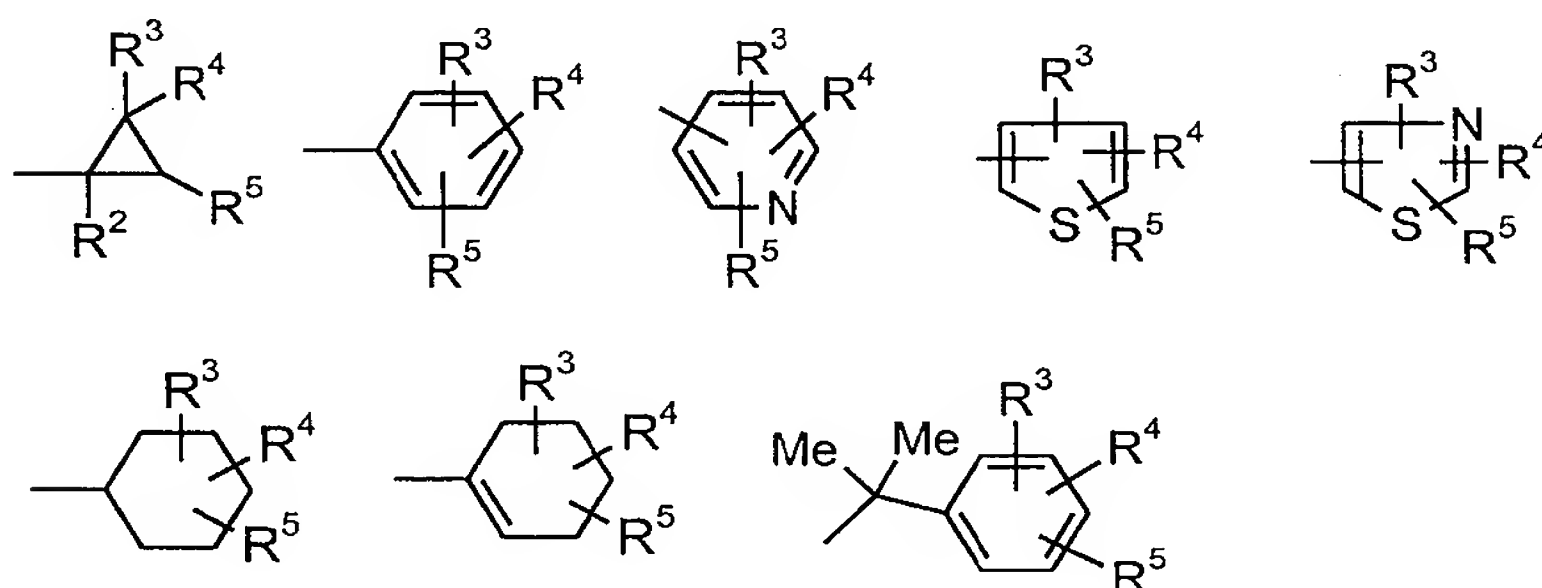


wherein

- 5 Z is pyridyl, optionally substituted with up to four groups independently selected from the group consisting of Cl, F, methyl, halomethyl, methoxy, halomethoxy, and methylthio;

one of X and Y is H, lower alkyl, haloalkyl, lower  
10 alkenyl, lower alkynyl, alkoxyalkyl, phenyl, or substituted phenyl;

the other of X and Y is a group selected from

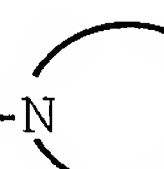
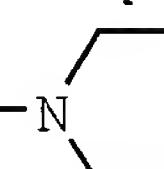







wherein


- 15 R<sup>2</sup> is halo, lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, or alkoxyalkyl;

R<sup>3</sup> is selected from H, halo, lower alkyl, (C<sub>7</sub>-C<sub>21</sub>) straight chain alkyl, hydroxy, lower alkoxy, haloalkyl, haloalkoxy, alkoxyalkyl, alkoxyalkoxy, lower alkenyl,  
20 haloalkenyl, CN, NO<sub>2</sub>, CO<sub>2</sub>R<sup>6</sup>, CON(R<sup>6</sup>)<sub>2</sub>, (C<sub>3</sub>-C<sub>6</sub>) cycloalkyl, S(O)<sub>m</sub>R<sup>6</sup>, SCN, pyridyl, pyridyloxy, substituted pyridyl, substituted pyridyloxy, phenoxy, substituted phenoxy, isoxazolyl, substituted isoxazolyl, naphthyl, substituted

naphthyl, phenyl, substituted phenyl,  $-(CH_2)_nR^6$ ,  $-CH=CHR^6$ ,  
 $-C\equiv CR^6$ ,  $-CH_2OR^6$ ,  $-CH_2SR^6$ ,  $-CH_2NR^6R^6$ ,  $-OCH_2R^6$ ,  $-SCH_2R^6$ ,

$-NR^6CH_2R^6$ ,  $-NCH_3NH_2$ ,  $-N$    $(CH_2)_p$ ,  $-N$    $O$ ,  $-NR^6$    $R^6$ ,

$-NR^6$    $OR^6$ ,  $-NR^6$    $N(R^6)$ ,  $-N(OR^6)$    $R^6$ ,  $-C(=O)H$ ,  $-O$    $R^6$ , or

5  $-S$    $R^6$ ;

$R^4$  and  $R^5$  are independently H, halo, lower alkyl,  
 lower alkoxy, haloalkyl, haloalkoxy, CN,  $NO_2$ ,  $CO_2R^6$ ,  
 $CON(R^6)_2$ , or  $S(O)_m$  alkyl, or

10  $R^4$  and  $R^5$  form a 5 or 6 member saturated or  
 unsaturated carbocyclic ring which may be substituted by  
 1 or 2 halo, lower alkyl, lower alkoxy or haloalkyl  
 groups;

$R^6$  is H, lower alkyl, haloalkyl, lower alkenyl,  
 lower alkynyl, phenyl, or substituted phenyl;

15 m is 0, 1, or 2; and

n is 1 or 2;

p is an integer from 2 to 6;

or a phytolegically acceptable acid addition salt or N-  
 oxide thereof.

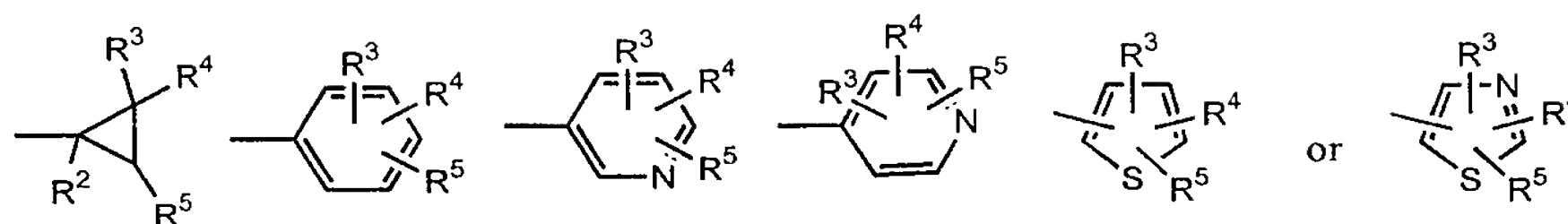
20

2. A compound of claim 1

wherein

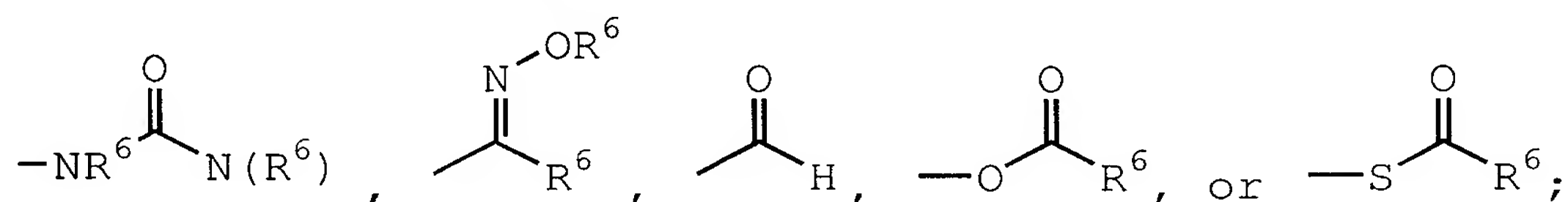
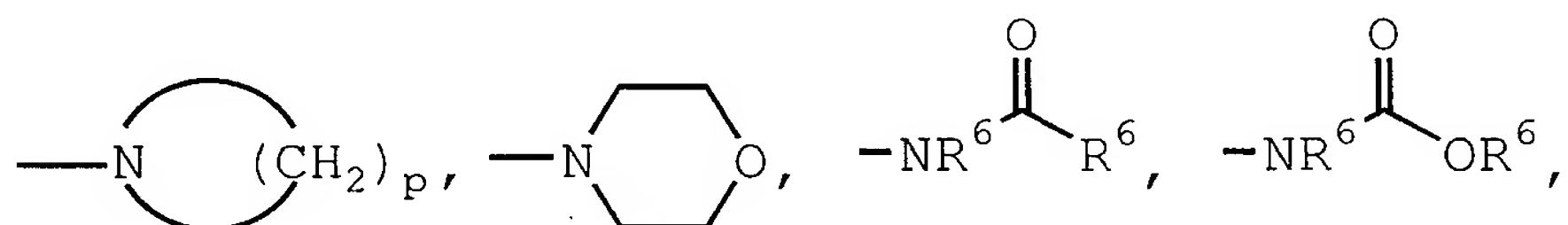
X is lower alkyl, haloalkyl, lower alkenyl, lower  
 alkynyl, or alkoxyalkyl;

25 Y is a group selected from



wherein

$R^3$  is selected from H, halo, lower alkyl, (C<sub>7</sub>-C<sub>21</sub>) straight chain alkyl, hydroxy, lower alkoxy, haloalkyl, haloalkoxy, alkoxyalkyl, alkoxyalkoxy, lower alkenyl, haloalkenyl, CN, NO<sub>2</sub>, CO<sub>2</sub>R<sup>6</sup>, CON(R<sup>6</sup>)<sub>2</sub>, (C<sub>3</sub>-C<sub>6</sub>) cycloalkyl, S(O)<sub>m</sub>R<sup>6</sup>, SCN, pyridyl, substituted pyridyl, isoxazolyl, substituted isoxazolyl, naphthyl, substituted naphthyl, phenyl, substituted phenyl, -(CH<sub>2</sub>)<sub>n</sub>R<sup>6</sup>, -CH=CHR<sup>6</sup>, -C≡CR<sup>6</sup>, -CH<sub>2</sub>OR<sup>6</sup>, -CH<sub>2</sub>SR<sup>6</sup>, -CH<sub>2</sub>NR<sup>6</sup>R<sup>6</sup>, -OCH<sub>2</sub>R<sup>6</sup>, -SCH<sub>2</sub>R<sup>6</sup>, -NR<sup>6</sup>CH<sub>2</sub>R<sup>6</sup>,



and

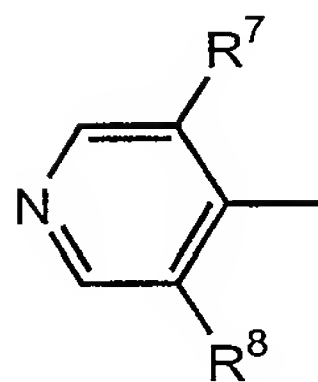
$R^4$  and  $R^5$  are independently H, halo, lower alkyl, lower alkoxy, haloalkyl, haloalkoxy, CN, CO<sub>2</sub>R<sup>6</sup>, CON(R<sup>6</sup>)<sub>2</sub>, or S(O)<sub>m</sub> alkyl, or

$R^4$  and  $R^5$  form a 5 or 6 member saturated or unsaturated carbocyclic ring which may be substituted by 1 or 2 halo, lower alkyl, lower alkoxy or haloalkyl groups.

3. A compound of claim 1 wherein Z is a 4-pyridyl group optionally substituted with up to four groups independently selected from Cl, F, methyl, halomethyl, methoxy, and halomethoxy.



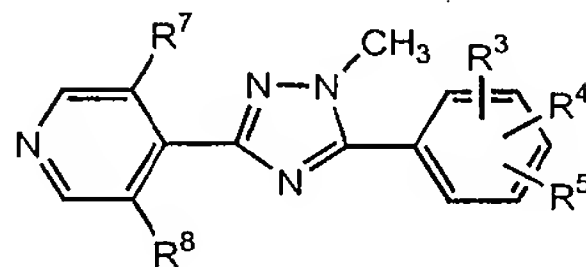
4. A compound of claim 2 wherein Z is a 4-pyridyl group of the formula



wherein  $R^7$  and  $R^8$  are independently H, Cl, F, methyl,  
5 halomethyl, methoxy, or halomethoxy.

5. A compound of claim 4 wherein  $R^7$  and  $R^8$  are independently F or Cl.

10 6. A compound of claim 1 having the formula



wherein

$R^7$  and  $R^8$  are independently F or Cl; and

15  $R^3$ ,  $R^4$ , and  $R^5$  are independently H, Cl, Br, methyl, halomethyl, methoxy, or halomethoxy.

7. The compound of claim 1 which is 3-(2,6-dichloro-4-pyridyl)-5-(4-chlorophenyl)-1-methyl[1,2,4]triazole.

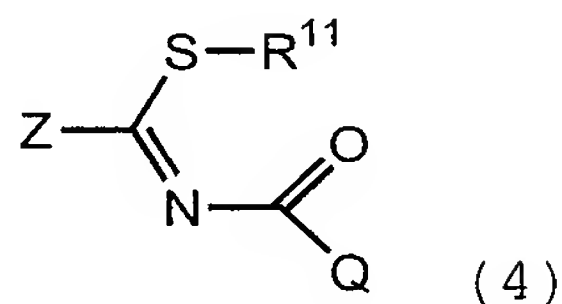
20 where  $R^3$ ,  $R^4$ , and  $R^5$  are as defined in claim 1.

8. A composition for controlling insects or mites which comprises a compound of any one of claims 1 to 7 in combination with a phytologically-acceptable carrier.

25 9. A method of controlling insects or mites which

comprises applying to a locus where control is desired an insect- or mite-inactivating amount of a compound of any one of claims 1 to 7.

5 10. A compound of the formula

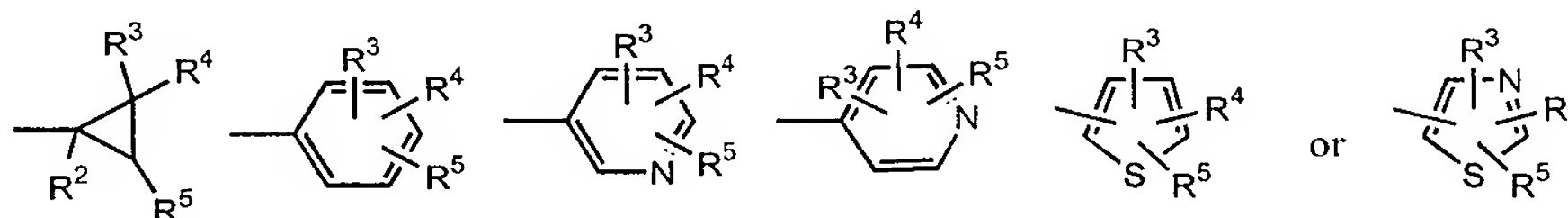


wherein

Z is 2-pyridyl, 3-pyridyl, or 4-pyridyl, optionally substituted with up to four groups independently selected from the group consisting of Cl, F, methyl, halomethyl, methoxy, and halomethoxy;

R<sup>11</sup> is lower alkyl; and

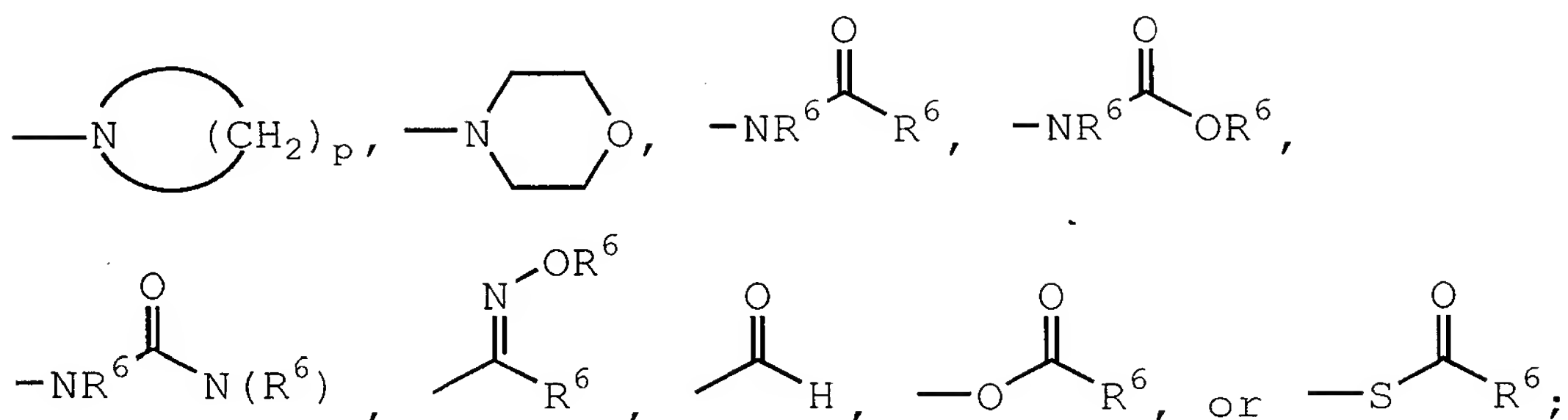
Q is a group selected from



15 wherein

R<sup>2</sup> is lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, or alkoxyalkyl;

R<sup>3</sup> is selected from H, halo, lower alkyl, (C<sub>7</sub>-C<sub>21</sub>) straight chain alkyl, hydroxy, lower alkoxy, haloalkyl, haloalkoxy, alkoxyalkyl, alkoxyalkoxy, lower alkenyl, haloalkenyl, CN, NO<sub>2</sub>, CO<sub>2</sub>R<sup>6</sup>, CON(R<sup>6</sup>)<sub>2</sub>, (C<sub>3</sub>-C<sub>6</sub>) cycloalkyl, S(O)<sub>m</sub>R<sup>6</sup>, SCN, pyridyl, substituted pyridyl, isoxazolyl, substituted isoxazolyl, naphthyl, substituted naphthyl, phenyl, substituted phenyl, -(CH<sub>2</sub>)<sub>n</sub>R<sup>6</sup>, -CH=CHR<sup>6</sup>, -C≡CR<sup>6</sup>, -CH<sub>2</sub>OR<sup>6</sup>, -CH<sub>2</sub>SR<sup>6</sup>, -CH<sub>2</sub>NR<sup>6</sup>R<sup>6</sup>, -OCH<sub>2</sub>R<sup>6</sup>, -SCH<sub>2</sub>R<sup>6</sup>, -NR<sup>6</sup>CH<sub>2</sub>R<sup>6</sup>,



$R^4$  and  $R^5$  are independently H, halo, lower alkyl, lower alkoxy, haloalkyl, haloalkoxy, CN,  $\text{CO}_2\text{R}^6$ ,  $\text{CON(R}^6\text{)}_2$ , or  $\text{S(O)}_m$  alkyl, or

$R^4$  and  $R^5$  form a 5 or 6 member saturated or unsaturated carbocyclic ring which may be substituted by 1 or 2 halo, lower alkyl, lower alkoxy or haloalkyl groups;

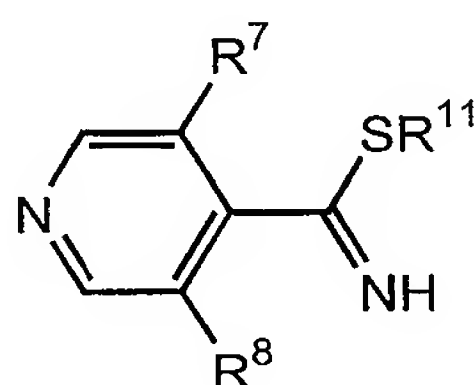
$R^6$  is H, lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, phenyl, or substituted phenyl;

$m$  is 0, 1, or 2; and

$n$  is 1 or 2;

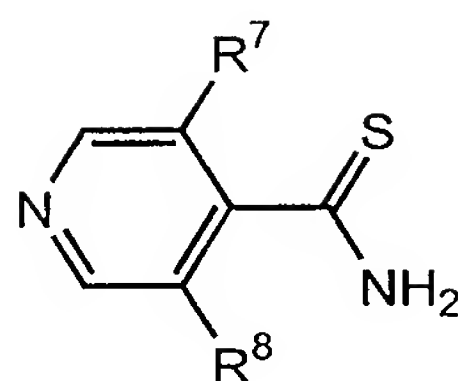
$p$  is an integer from 2 to 6.

11. A compound of the formula



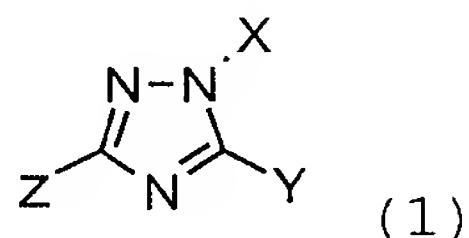
wherein  $R^7$  and  $R^8$  are both F or both Cl and  $R^{11}$  is lower alkyl, or an acid addition salt thereof.

12. A compound of the formula



wherein  $R^7$  and  $R^8$  are both F or both Cl.

13. A process for preparing compound of the formula (1)

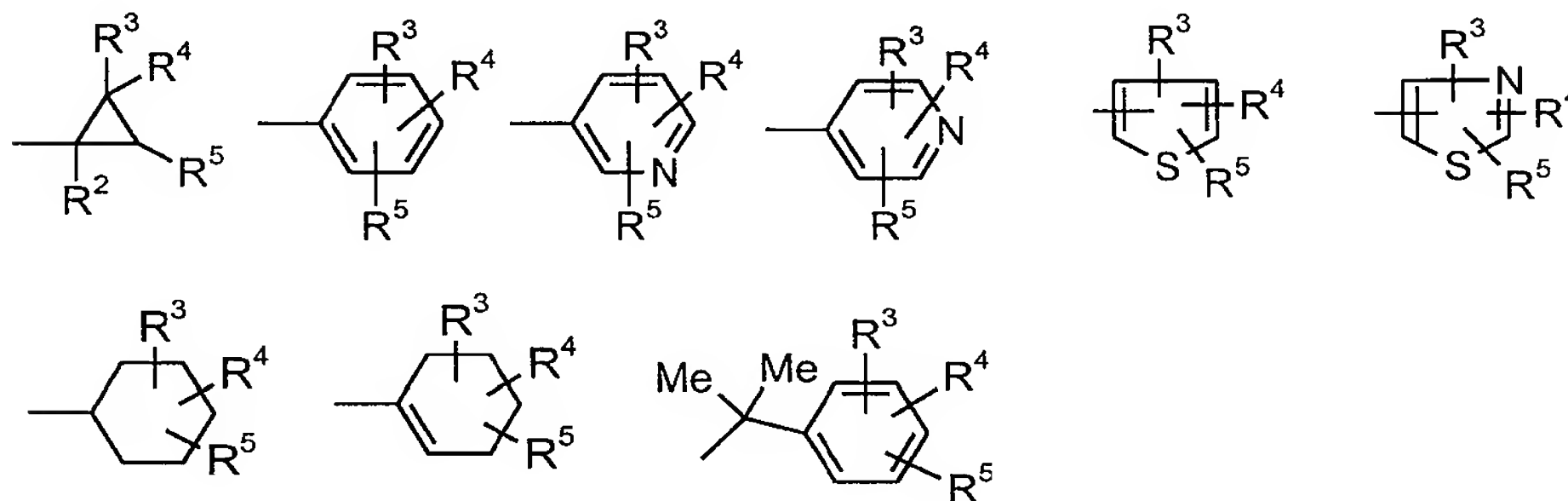


wherein

Z is pyridyl, optionally substituted with up to four groups independently selected from the group consisting of Cl, F, methyl, halomethyl, methoxy, and halomethoxy;

X is H, lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, alkoxyalkyl, phenyl, or substituted phenyl;

Y is a group selected from

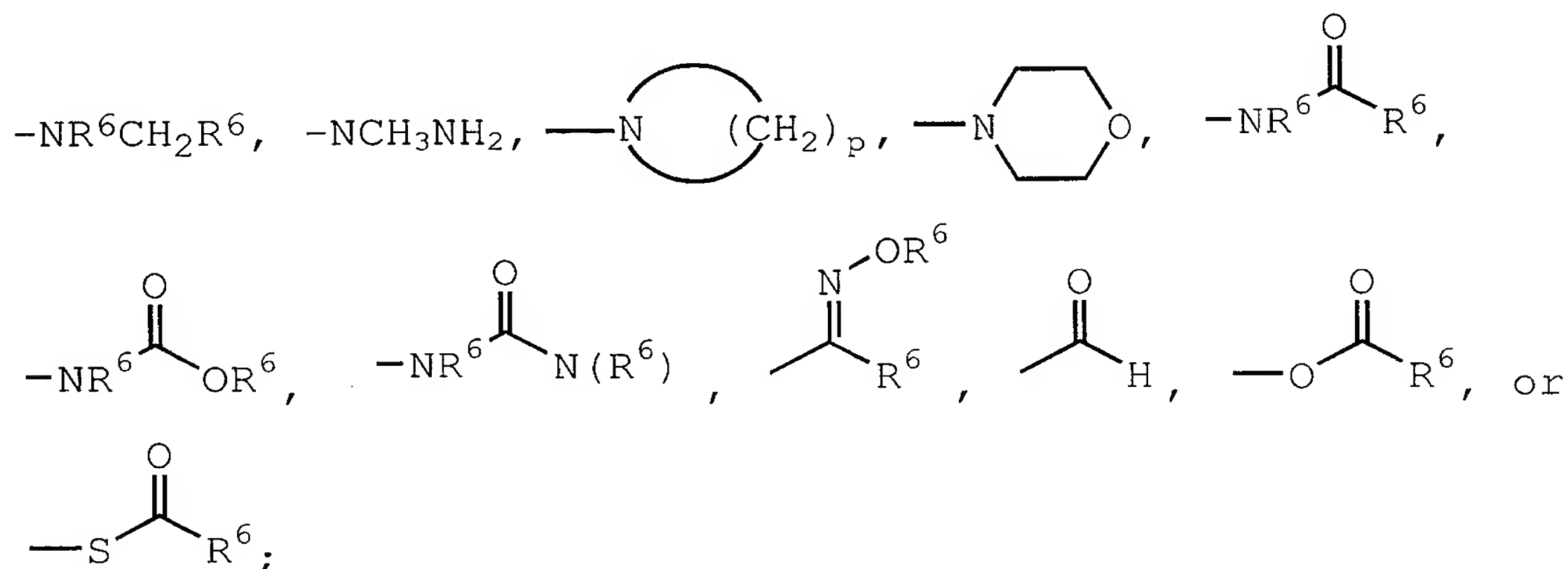


wherein

$R^2$  is lower alkyl, haloalkyl, lower alkenyl, lower alkynyl, or alkoxyalkyl;

$R^3$  is selected from H, halo, lower alkyl, (C<sub>7</sub>-C<sub>21</sub>) straight chain alkyl, hydroxy, lower alkoxy, haloalkyl, haloalkoxy, alkoxyalkyl, alkoxyalkoxy, lower alkenyl, haloalkenyl, CN, NO<sub>2</sub>, CO<sub>2</sub>R<sup>6</sup>, CON(R<sup>6</sup>)<sub>2</sub>, (C<sub>3</sub>-C<sub>6</sub>) cycloalkyl,

S(O)<sub>m</sub>R<sup>6</sup>, SCN, pyridyl, pyridyloxy, substituted pyridyl,  
 substituted pyridyloxy, phenoxy, substituted phenoxy,  
 isoxazolyl, substituted isoxazolyl, naphthyl, substituted  
 naphthyl, phenyl, substituted phenyl, -(CH<sub>2</sub>)<sub>n</sub>R<sup>6</sup>, -CH=CHR<sup>6</sup>,  
 5 -C≡CR<sup>6</sup>, -CH<sub>2</sub>OR<sup>6</sup>, -CH<sub>2</sub>SR<sup>6</sup>, -CH<sub>2</sub>NR<sup>6</sup>R<sup>6</sup>, -OCH<sub>2</sub>R<sup>6</sup>, -SCH<sub>2</sub>R<sup>6</sup>,



R<sup>4</sup> and R<sup>5</sup> are independently H, halo, lower alkyl,  
 10 lower alkoxy, haloalkyl, haloalkoxy, CN, NO<sub>2</sub>, CO<sub>2</sub>R<sup>6</sup>,  
 CON(R<sup>6</sup>)<sub>2</sub>, or S(O)<sub>m</sub> alkyl, or

R<sup>4</sup> and R<sup>5</sup> form a 5 or 6 member saturated or  
 unsaturated carbocyclic ring which may be substituted by  
 1 or 2 halo, lower alkyl, lower alkoxy or haloalkyl  
 15 groups;

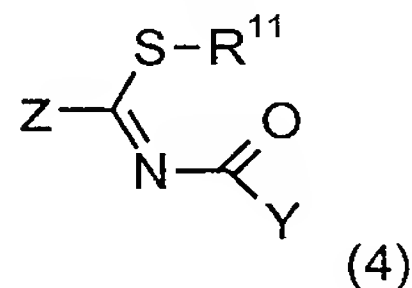
R<sup>6</sup> is H, lower alkyl, haloalkyl, lower alkenyl,  
 lower alkynyl, phenyl, or substituted phenyl;

m is 0, 1, or 2; and

n is 1 or 2;

20 p is an integer from 2 to 6;

which comprises reacting a compound of formula (4)



where R<sup>11</sup> is lower alkyl and Z and Y are as defined above,  
 with methylhydrazine.

## INTERNATIONAL SEARCH REPORT

Inte. onal Application No

PCT/US 99/24751

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C07D401/04 C07D401/14 C07D409/14 C07D417/14 C07D213/83  
A01N43/653

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C07D A01N

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the International search (name of data base and, where practical, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 185 256 A (HOFFMANN LA ROCHE) 25 June 1986 (1986-06-25) column 28, line 47 - line 59; claims 1,8; example 2	1-4
X	EP 0 217 552 A (SUMITOMO CHEMICAL CO) 8 April 1987 (1987-04-08) claims; examples	1,2
X	FR 2 653 432 A (RHONE POULENC AGROCHIMIE) 26 April 1991 (1991-04-26) claim 1; examples	1,2
X	EP 0 181 282 A (CIBA GEIGY AG) 14 May 1986 (1986-05-14) example 3	1,2
	-/-	

☒ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

## \* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&amp;" document member of the same patent family

Date of the actual completion of the international search

28 December 1999

Date of mailing of the international search report

11/01/2000

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Authorized officer

De Jong, B

# INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 99/24751

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 4 011 218 A (BALDWIN JOHN J ET AL) 8 March 1977 (1977-03-08) claims; examples	1-4
X	US 3 984 558 A (BALDWIN JOHN J ET AL) 5 October 1976 (1976-10-05) example 1	1-4
X	US 3 963 731 A (NOVELLO FREDERICK C ET AL) 15 June 1976 (1976-06-15) examples	1-4
X	DE 23 58 011 A (LEPETIT SPA) 11 July 1974 (1974-07-11) claim 1; examples 1-4,13,14	1-4
X	DE 22 58 036 A (GOEDECKE AG) 30 May 1974 (1974-05-30) example 82	1-4
X	FR 1 238 943 A (BASF) 7 December 1960 (1960-12-07) the whole document	1-4
A	EP 0 559 363 A (KUMIAI CHEMICAL INDUSTRY CO ;IHARA CHEMICAL IND CO (JP)) 8 September 1993 (1993-09-08) claim 1; examples	1

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Information on patent family members

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Information on patent family members

International Application No

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# INTERNATIONAL SEARCH REPORT

International application No.

PCT/US 99/ 24751

## Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:  
because they relate to subject matter not required to be searched by this Authority, namely:
2. ☒ Claims Nos.:  
because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:  
see FURTHER INFORMATION sheet PCT/ISA/210
3. ☐ Claims Nos.:  
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

## Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this International application, as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.
- ☐ No protest accompanied the payment of additional search fees.

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

Continuation of Box I.2

Claims Nos.: not applicable

The search for claims 1-4 revealed such a large number of particularly relevant documents, in particular with regard to novelty, that the drafting of a comprehensive European Search Report is not feasible. The cited documents are considered as to form a representative sample of the revealed documents, duly taking into account their relevance with respect to the subject-matter as illustrated by the examples.

# INTERNATIONAL SEARCH REPORT

Information on patent family members

Int'l Application No

PCT/US 99/24751

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
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